# Stereoselective and Catalytic Access to $\beta$ -Enaminones: An Entry to Pyrimidines

Eric Gayon,<sup>†</sup> Monika Szymczyk,<sup>†</sup> Hélène Gérard,<sup>\*,‡</sup> Emmanuel Vrancken,<sup>\*,†</sup> and Jean-Marc Campagne<sup>\*,†</sup>

<sup>†</sup>Institut Charles Gerhardt UMR 5253 CNRS-UM2-UM1-ENSCM, 8 rue de l'Ecole Normale, 34296 Montpellier Cedex 5, France <sup>‡</sup>Laboratoire de Chimie Théorique, UMR 7616, UPMC—Université Paris 06, CNRS case 137, 4 Place Jussieu, 75262 Paris Cedex 05, France

**Supporting Information** 

**ABSTRACT:** We describe herein a highly stereoselective access to Cbz-protected  $\beta$ -enaminones 2 based on the NaOH catalyzed rearrangement of propargylic hydroxylamines 1. The synthetic potential of these  $\beta$ -enaminones is illustrated in an original synthesis of pyrimidines.



# INTRODUCTION

 $\beta$ -Enaminones (vinylogous amides) are versatile compounds,<sup>1</sup> used for their pharmacological properties<sup>2</sup> and as building blocks for heterocyclic<sup>3</sup> and natural product<sup>4</sup> syntheses. Since their biological and chemical properties depend on the alkene stereochemistry, stereoselective syntheses of  $\beta$ -enaminones are thus highly desirable.<sup>5</sup> During initial experiments aimed at developing an interrupted access to trisubstituted isoxazoles,<sup>6</sup> we observed that a propargylic hydroxylamine (Ar = *p*-tol, R = *n*-Bu) was able, under basic conditions, to rearrange to give the Cbzprotected enaminone as a single (*Z*) diastereomer (eq 1, see below for theoretical justification):



Puzzled by this stereoselective isomerization, we were keen to investigate this transformation and to use the enaminones 2 as building blocks for the synthesis of heterocyclic structures such as pyrimidines. Pyrimidines are ubiquitous heterocyclic moieties present in natural products, drugs and in functional materials." Efficient strategies have been developed mainly on the basis of N-C-N condensations, i.e., the addition of guanidines or presynthetized amidines salts to 1,3-diketones or derivatives thereof (such as ynones<sup>7e</sup>) (Scheme 1, eq 1). The presence of a nitrogen atom on enaminones 2 could be advantageous to develop alternative synthetic strategies since it allows for differentiation of the 2 nitrogen groups (which could, for example, be useful for isotopic labeling strategies). Moreover, such a cyclocondensation would use the readily commercially available and stable class of carboxamides, which grants a more convenient access to these heterocycles.8

Herein, we describe the catalytic isomerization of readily available propargylic hydroxylamines  $1^{9,10}$  to (Z)- $\beta$ -enaminones **2** and we propose a reaction pathway based on DFT calculations for this unanticipated isomerization. The further transformation

# Scheme 1. Synthesis of Pyrimidines



of 2 into 2,4,6-trisubstituted pyrimidines is also reported (Scheme 1, eq 2).

## RESULTS AND DISCUSSION

The starting propargylic hydroxylamines 1 are easily obtained by FeCl<sub>3</sub>-catalyzed direct nucleophilic substitution of the propargylic hydroxyl group by the commercially available Cbz-NHOH under reaction conditions developed by Zhan (FeCl<sub>3</sub> 5 mol %, DCM, 60 °C, 1 h).<sup>10</sup> First, **1a** was chosen as model substrate and its isomerization under various basic conditions was investigated (Table 1).

Starting from the previously described conditions (4.0 equiv of  $K_2CO_3$  in acetonitrile at 50 °C for 20 h),<sup>6</sup> the influence of the base was first tested: whereas no reaction is observed with NEt<sub>3</sub> (entry 2), the conversion is complete after 1 h using sodium hydroxide (4 equiv), leading to **2a** in 28% isolated yield along with unidentified byproduct (entry 3). The (*Z*) geometry of the double bond was assigned by <sup>1</sup>H NMR chemical shift correlation. Indeed, the <sup>1</sup>H NMR shift of the proton attached to the nitrogen atom of our enaminones **2** is in good agreement with previously described (*Z*)- $\beta$ -enaminones ( $\delta_{N-H}$ : ca. 11 ppm).<sup>11</sup> In sharp contrast, in enecarbamates, which can be seen as analogs where

Received: August 22, 2012 Published: September 24, 2012 Table 1. Isomerization of Propargylic Hydroxylamine 1a under Various Basic Conditions

	Cbz OH		Acetonitrile 50 °C	<sup>z</sup> NH O h <i>n-Bu</i> <b>2a (</b> <i>Z</i> / <i>E</i> > 98:2)
	base	equiv	reaction time (h)	2a isolated yield (%)
1	K <sub>2</sub> CO <sub>3</sub>	4	20	71
2	NEt <sub>3</sub>	4	20	NR
3	NaOH	4	1	28
4	NaOH	1	1	75
5	NaOH	0.1	1	86
6	$K_2CO_3$	0.1	8	59

the carbonyl is not present, a lower shift is observed ( $\delta_{\rm N-H}$ : ca. 6 ppm).<sup>12</sup> This shielding effect observed in the case of enaminones is characteristic of the existence of a H bonding, which is in good agreement with the proposed (Z) geometry. Moreover, the structure of 2 was further confirmed by nOesy experiments (see Supporting Information for details).

Anticipating that an excess of base might be problematic, the reaction was carried out with only 1 equiv of NaOH. Indeed, 2a was isolated in 75% as a single (Z) isomer (entry 4). Considering that the isomerization process should lead to an anionic product, we assumed that a catalytic amount of base might be sufficient. Thus, the use of 10 mol % of NaOH was finally attempted and, gratifyingly, 2a was isolated in an improved 86% yield (entry 5). In a control experiment (0.1 equiv of  $K_2CO_3$  in acetonitrile at 50 °C), compound 2a was isolated in 59% yield and the reaction required 8 h for completion (entry 6). It is worth noting that the presence of water is required to ensure good conversions (see the mechanistic study for explanations and in particular about the role of water). Consequently, our reactions are run in nondistilled acetonitrile<sup>13</sup> and in an open-air flask, which makes this methodology quite practical.

We next studied the scope and limitations of this isomerization. Using our optimized conditions (Scheme 2), the influence of the aryl substituents at the propargylic position was first investigated. Starting from propargylic hydroxylamines bearing electron-withdrawing or electron-donating groups, at the para or meta position, Cbz-protected enaminones 2a-i were obtained in good to excellent yields. Heteroaromatics, such as thiophene, are also well tolerated in this reaction (2j). Some limitations appear when using more hindered substrates: compare for example 1-naphtyl vs 2-naphthyl moieties (2k and 21) or o-Me vs o-F substituents (2m and 2n). Taken together, these results show that the presence of a sterically hindered group close to the propargylic position has a detrimental effect on the yield. Mechanistic implications of this observation are discussed thereafter.

Moving to a TMS group, and under classical conditions, the attempted enaminone 20 could not be observed. After some optimization, and using 1 equiv of K<sub>2</sub>CO<sub>3</sub> at room temperature in a methanol/water mixture, compound 4 could finally be isolated in 68% yield (Scheme 3). Of note, this product is isolated as a single Z isomer and results from a one-pot transformation including three elemental steps: isomerization, desilylation and Cbz deprotection. It is not clear when the Cbz deprotection occurs in this sequence; nevertheless, knowing the stability of acylsilane under hydrolysis conditions,14 the formation of a formyl group in the final product indicates that the isomerization step takes place after the TMS removal. Clearly, the presence of a

Scheme 2. Scope of the Isomerization of Propargylic Hydroxylamines: Influence of the Aryl Group at the



<sup>a</sup>General conditions: Propargylic hydroxylamine (0.2 mmol), NaOH (0.02 mmol, 10 mol %), CH<sub>3</sub>CN (2 mL, 0.1 M), 50 °C, 1 h. <sup>b</sup>Isolated yields.

Scheme 3. Isomerization of a Propargylic Hydroxylamine Substituted by a TMS Group at the Acetylenic Position



TMS at the acetylenic position inhibits the isomerization, whereas terminal alkynes are suitable reagents for this transformation.

Next, substrates bearing different alkyl or aryl substituents on the acetylenic position were tested. With Me and *i*-Pr alkyl groups, the corresponding enaminones 2p,q were isolated in 75 and 63% yields, respectively (Scheme 4, eq 1 and 2). In contrast, the expected enaminone 2r could only be observed as traces in the crude material when a bulkier t-Butyl group stands at the acetylenic position. Along with unreacted starting material 1r (recovered in 46% yield), the major product was the imine 3r isolated in 41% yield (Scheme 4, eq 3). With aryl groups (Ph and *p*-Tol), the expected enaminones were obtained in 51% (2s) and 53% (2t) yields along with the corresponding imines 3s and 3t in 18 and 14% yields, respectively (Scheme 4, eq 4).

The detection of imines 3r-t provides some insight on the reaction mechanism, which thus appears to be initiated by the  $\beta$ elimination of the OH group. This transformation is rare in

Article

Scheme 4. Isomerization of Propargylic Hydroxylamines: Influence of Alkyl or Phenyl Substituent at the Acetylenic Position



literature,<sup>15</sup> and we were particularly puzzled that it could proceed at relatively low temperature in basic conditions. Moreover, the synthesis of  $\beta$ -enaminones from imines was not documented. We thus decided to elucidate the mechanism of this isomerization, and starting from the observation that imines were formed in the reaction mixture, a mechanism model was sought using a computational approach (DFT based; see Theoretical Section for details). A complete catalytic path was found, including the succession of steps described in Scheme 5.





The full reaction sequence was found to take place in the anionic, deprotonated form, in the presence of water. It is starting from the deprotonated aminol 1u' that is readily formed upon geometry optimization when HO<sup>-</sup> is associated to 1u (no cation is considered since ion pairs are supposed to exclusively exist as solvent separated ion pairs (SSIP) in such a polar solvent; see Theoretical Section). The following steps are observed, where the deprotonated forms are noted with prime:

- HO<sup>-</sup> elimination triggered by deprotonation at the propargylic position to form the transient conjugated imine 3u.
- (2) Addition of  $HO^-$  to 3u yielding an allenol 5u'.
- (3) Keto-enol tautomerization of  $5\mathbf{u}'$  to conduct to deprotonated enaminone  $2\mathbf{u}'$ .
- (4) Proton exchange between the reactant 1u and the anionic product 2u'.

Step one was found to be a syn elimination as, even though the anti mechanism exhibits a low energy barrier (less than 0.5 kcal  $mol^{-1}$ ), it cannot take place in the deprotonated aminol form 1u', which is found to be preferred by 22.0 kcal mol<sup>-1</sup> over the structure leading to the anti process. Step two is associated with the addition of the HO<sup>-</sup> at the position 4 of the imine 3u to yield allenol 5u', but no reprotonation of the amide is observed. As for the starting aminol 1u, when including an explicit water molecule in the modeled system, the most stable form is associated to a delocalized organic anion and a water molecule. An anionic structure is also predominant after the tautomerization, step 3, as the protonation at the central carbon of allenol 5u' and the deprotonation of the OH group are simultaneous and mediated through an H-bonded water molecule. Even though the proposed reaction step already leads to the right Z/E isomer of 2u', isomerization within this structure is easy, as rotation between the Z and E isomers exhibit an energy barrier of only 5.7 kcal mol<sup>-1</sup>. The last step of the reaction mechanism is a reprotonation of 2u' by the starting material (1u in Scheme 5) to yield the final product 2u, which is found to be exothermic by 1.3 kcal/mol<sup>-1</sup> and thus fully equilibrated. The overall energy profile is reported in Figure 1.



**Figure 1.** Energy profile (energies in italics are given in kcal/mol) of the HO<sup>-</sup> catalyzed isomerization process. Numbers are introduced in Scheme 5.

This reaction profile is found to be in complete agreement with experimental observations. The first step of this reaction path is supported by the lack of reactivity when a weak base such as NEt<sub>3</sub> ( $pK_a < 10$ ) is used (see Table 1, entry 2). We also observed a drastic drop of the conversion when the reaction is conducted in strictly anhydrous conditions under argon atmosphere, which confirms the crucial role of water in the catalytic process.<sup>13</sup> In two additional control experiments, we confirmed this hypothesis by submitting propargylic hydroxylamines 1v,w to our standard conditions (Scheme 6, eq 1). As expected, no reaction occurs,

# Scheme 6. Complementary Experiments Performed to Support the Mechanistic Model



and the starting material is fully recovered. In this case, because of the replacement of the aryl by an alkyl group in the propargylic position, the propargylic proton is not acidic enough to allow the first deprotonation step. As evidence for the second step of our proposed mechanism, the formation of imines is observed when a sterically hindered *t*-Bu substituent stands in the acetylenic position (Scheme 4, eq 3) or when the intermediate imine is fully conjugated (Scheme 4, eq 4), i.e., when the 1,4 addition step is disfavored. Moreover, when imine  $3a^{16}$  was subjected to a stoichiometric amount of NaOH, the corresponding enaminone 2a was obtained in almost quantitative yield, which shows that imine is a reaction intermediate (Scheme 6, eq 2).

To complete the theoretical study, the stereoselectivity of the final product was examined computationally through the relative stability of the final product 2u: indeed, the low isomerization barrier between Z and E forms found in 2u' associated with the equilibrium of reprotonation of 2u' to 2u let us conclude that both isomers are in thermodynamic equilibrium in the final product. Geometry optimization was thus carried out on the products, using increasing size models (see Figure 2): model A, where only H-bond can stabilize the structure, model B where Ph and Me substituent effects are taken into account, and 2u, which adds the C(O)OMe group at the nitrogen atom, with or without the effect of a continuous representation of the solvent. An anionic model 2u' deprotonated at the nitrogen and thus unable to stabilize through H-bonding is also examined. Results are given in Table 2. Whatever the model, a Z stereoisomer is systematically favored over the *E* one and is evaluated to 6.8 kcal  $mol^{-1}$  in the case of the model **2u** (4.1 kcal/mol<sup>-1</sup> in presence of a continuum model of the solvent). In absence of substituent effects, and thus when only the H-bond is at stake, the preference is only 4.0 kcal mol<sup>-1</sup>, in good agreement with the results published for similar species.<sup>17</sup> In absence of H-bonding, in the anionic form, the preference for the Z form is maintained, with a 3.0 kcal mol<sup>-1</sup> value. It is thus found that the preference for the Z isomer is due both to substituent effects (about 3 kcal/mol<sup>-1</sup>, steric hindrance of the Ph and Me in the *E* isomer, see Figure 2,

Table 2. Energies of the Various Stereo isomers Relative to the Most Stable One for Each Model "  $\,$ 

		<i>E</i> -trans	E-cis	Z-trans	Z-cis
	model A	4.0	5.1	6.8	0.0
	model B	9.5	9.9	8.3	0.0
	2u	6.9	6.8	7.2	0.0
	$2u_{solv}$	4.1	5.3	4.4	0.0
	2u'	3.0	3.5	0.0	5.6
T.,.	$ r_{aa} /m_{a} ^{-1}$	The	<b>1</b> " - <b>6 J</b> -	f	

"In kcal/mol $^{-1}$ . The subscript "solv" stands for results in a continuum model for the solvent.

model B) and stabilization by an H-bond (about  $4 \text{ kcal/mol}^{-1}$ ), the latest being weakened (by about 2.7 kcal/mol<sup>-1</sup>) in presence of a polar solvent. This double effect sheds a new light on the commonly accepted picture of stabilization by intramolecular H-bonding.<sup>17</sup>

In the first part of our work, we explored the isomerization of propargylic hydroxylamine into  $\beta$ -enaminones and built a plausible mechanism for this original transformation. We next decided to explore the synthetic potential of the Cbz-protected enaminones **2**. We reasoned that enaminone could represent interesting building block for the synthesis of heterocyclic compounds such as pyrimidines. Indeed, the presence of a geometrically defined double bond as well as a nucleophilic nitrogen and an electrophilic carbonyl groups should allow the development of an alternative synthesis based on the use of a complementary electrophile/nucleophile partner such as carboxamide, in a classical ionic cyclocondensation process. To test this hypothesis, the reactivity of enaminone **2a** with benzamide was screened under various conditions (Table 3).

Table 3. Formation of Pyrimidine 6 from Enaminone 2a andBenzamide: Optimization Study





Figure 2. Models examined for relative stability of the Z and E isomers.

Scheme 7. Attempts of Cyclocondensation Starting from Benzonitrile or Cbz-Deprotected 8







Under acidic conditions (entry 1), the expected pyrimidine 6 was observed along with the hydrolysis product 7 (1:1 ratio along with unidentified products). In contrast, in the presence of MeONa (entry 2), pyrimidine 6 could be isolated in 30% yield. This low yield results from the competitive formation of the Cbz deprotected enaminone 8 (6 and 8 are in a 1:1 ratio in the crude mixture). To avoid the competitive Cbz deprotection, the use of bulkier bases such as DBU or *t*-BuOK was investigated. Whereas deceptive results were observed with DBU (entry 3), the use of 1 equiv of t-BuOK led to 6 in 51% yield, along with unreacted starting material 2a (35%) (entry 4). The amount of base is crucial, and 2 equiv of t-BuOK appeared to be necessary to gain full conversion and produce pyrimidine 6 in 80% yield (see entries 5 and 6). Unfortunately, all tries to set up an efficient onepot two-step access to pyrimidines from N-hydroxypropargylamines 1 failed.<sup>18</sup> Indeed, the isolated yield was lowered compared to the stepwise process due to the presence of deprotected enaminone.

Attempt to use a nitrile as coupling partner gave unsatisfying results: when enaminone 2a and benzonitrile were engaged in the previously optimized reaction conditions, pyrimidine 6 and deprotected enaminone 8 were obtained in 28 and 26% yield, respectively (Scheme 7, eq 1). Of note, when 8 was treated with benzamide, no trace of pyrimidine could be observed in the crude mixture, which highlights (i) that 8 is not an intermediate in the formation of 6 and (ii) the crucial role of the Cbz protecting group for the reaction process (Scheme 7, eq 2). Consequently, some insights into the mechanistic pathway can be brought forward. These two remarks give some insights to the mechanistic pathway of this transformation. If the exact course of the nucleophilic attack steps remains unclear, it seems reasonable to assume that the removal of the Cbz group should occur during the aromatization step, as recently proposed by Kuninobu and Takai for the formation of multisubstituted pyridines.<sup>19</sup>

For practical reasons, these optimization studies have been carried out on small scale (0.2 mmol), using sealed vials. Consequently, these optimized conditions were next tried on a larger scale, using classical glassware, and gratifyingly, pyrimidine **6** was isolated in 81% yield (3.05 mmol, 880 mg, Scheme 8).

With our optimized conditions (2.0 equiv of tBuOK in refluxing toluene), the scope and limitations of this pyrimidine synthesis were surveyed. As illustrated in Scheme 9, differently substituted enaminones lead, in the presence of carboxamides, to the corresponding pyrimidines 6, 9-22 in moderate to good yields. Starting from benzamide, pyrimidines 6, 9-16 were obtained in 52-89% yields. Interestingly, the presence of electron-donating groups at the meta or para position has a positive influence on the yield. Nevertheless, halogen substituents are tolerated, and notably, the presence of a bromine atom in compound 13 allows for further functionalization. The nature of the alkyl substituent (primary or secondary) at the carbonyl group  $(R^2)$  has little influence (compounds 17, 18), whereas a slight decrease in yield is observed when R<sup>2</sup> is a phenyl group (54%, compound 19). The scope has been successfully extended to heteroaromatic (20) and para-nitro and -bromo substituted arenes (21 and 22, respectively).

2-Vinyl substituted pyrimidines are particularly interesting substrates,<sup>20</sup> usually obtained via cross-coupling reaction from the corresponding 2-chloro pyrimidines. Starting from acryl-amide, pyrimidines **23–25** have been obtained in 55, 51, and 46% yields, respectively (Scheme 10). Anticipating that these pyrimidines could be sensitive to polymerization, the same reactions were carried out in the presence of TEMPO (1 equiv) as a radical inhibitor. Under these conditions, pyrimidines **23–25** were obtained in improved 72, 69, and 68% yields, respectively. Finally, simple alkyl carboxamides, such as acetamide, can be used as partners for this reaction, leading to **26** with a satisfying 72% isolated yield. Starting from formamide, the use of molecular sieves proved to be detrimental, but in its absence, 4,6-disubstituted pyrimidine **19** was finally obtained in 49% yield.

Scheme 9. Synthesis of Pyrimidines from Aromatic and Heteroaromatic Carboxamides: Scope and Limitations<sup>a</sup>



<sup>*a*</sup>General conditions: vinylogous amide (0.2 mmol), amide (0.3 mmol, 1.5 equiv), *t*-BuOK (0.4 mmol, 2 equiv), 4 Å MS (70 mg), PhCH<sub>3</sub> (2 mL, 0.1 M), 130 °C, 1 h.

#### CONCLUSIONS

In conclusion, we described a catalytic and stereoselective isomerization of readily available propargylic hydroxylamines 1 leading  $\beta$ -enaminones 2 under simple and efficient conditions (NaOH 10 mol %, 1 h). This unanticipated transformation was rationalized with the help of an intertwined theoretical/ experimental study that allows determining each step of the catalytic pathway, i.e.,  $\beta$ -elimination, 1,4-addition and tautomerization. The first  $\beta$ -elimination step, which leads to the N–O bond breaking, is of particular interest since it occurs via an original water-assisted six-membered syn elimination process. Moreover, the model generally admitted to explain the preference for Z form of  $\beta$ -enaminone has been revisited, and besides the internal stabilizing H-bonding, the substituents effect has been highlighted. The synthetic potential of Cbz-protected  $\beta$ -enaminones 2 was illustrated through the development of a novel access to pyrimidines with stable, commercially available and inexpensive carboxylic amides.

# THEORETICAL SECTION

Full geometry optimizations were systematically conducted with no symmetry restraints using the Gaussian 09 program<sup>21</sup> within the framework of the Density Functional Theory (DFT) using the dispersion corrected B97D<sup>22</sup> exchange-correlation functional (which was shown to give good results on similar questions<sup>23</sup>) and the 6-31+G\*\* basis set for all atoms. Frequencies were evaluated within the harmonic approximation and the nature of all minima was ensured by confirming the presence of no imaginary frequency. A model system **1u** was examined, derived from **1i** by replacing the Cbz C(O)OCH<sub>2</sub>Ph group by a simple C(O)OMe one, which is proposed to be a minor modification as (i) no major electronic effect as conjugation is supposed to be altered as non is expected for this group and (ii) the benzyl group is flexible enough to behave sterically like a methyl in this kind of compound. Solvation was taken into account using an implicit solvation model, and all geometry were fully optimized in presence of the

Scheme 10. Synthesis of Pyrimidines from Various Carboxamide Derivatives<sup>c</sup>



<sup>a</sup>Reaction conducted with addition of TEMPO (0.2 mmol, 1 equiv). <sup>b</sup>Reaction carried out without 4 Å MS, and starting enaminone **2a** was recovered in 26% yield. <sup>c</sup>General conditions: vinylogous amide (0.2 mmol), amide (0.3 mmol, 1.5 equiv), *t*-BuOK (0.4 mmol, 2 equiv), 4 Å MS (70 mg), PhCH<sub>3</sub> (2 mL, 0.1 M), 130 °C, 1 h.

continuum. For these computations, the PCM model was used within the default framework implemented in Gaussian 09,<sup>24</sup> except for atomic radii where the UFF values are used. The dielectric constant implemented for DMSO ( $\varepsilon_{\rm R}$  = 46.826) was used, a value slightly higher than that for acetonitrile ( $\varepsilon_{\rm R}$  = 35.688), in order to take into account the presence of non-negligible amount of water in the reaction media. The key intermediates were reoptimized in a model of acetonitrile in order to confirm the validity of this approach. Because of the high polarity of the solvent, NaOH is considered as a fully dissociated ion pair, and no Na<sup>+</sup> cation is included in the computational model.

#### EXPERIMENTAL SECTION

**General Considerations.** Unless otherwise stated, all commercial materials were used without further purification. Spherical 4 Å MS is used when needed. Chromatography was carried out on silica gel 60 A (35–70 mm). PMA stands for phosphomolybdic acid (ethanolic solution) and is used as TLC stain. Solvents (acetonitrile, toluene and methanol) were used without further purification. <sup>1</sup>H and <sup>13</sup>C NMR spectra were obtained on a 400 MHz NMR spectrometer and referenced to CDCl<sub>3</sub>. <sup>19</sup>F NMR spectra are calibrated using C<sub>6</sub>F<sub>6</sub> as an external standard. High resolution mass spectra (HRMS) were measured using electrospray ionization and were acquired using a Q-TOF analyzer. Propargylic alcohols have already been described and were easily obtained by the addition of the corresponding alkynyl lithium reagent (in anhydrous THF) to the corresponding aldehyde. <sup>9</sup>C,25

General Procedure A for the Preparation of Propargylic Hydroxylamines (1a–t). To a solution of the corresponding propargylic alcohol (1 equiv) in dichloromethane (C = 0.1 M) was added N-hydroxybenzenesulfonamide (1 equiv) and FeCl<sub>3</sub> (5 mol %, 0.05 equiv), and the mixture was refluxed for 90 min. After reaction completion (TLC monitoring), the mixture was concentrated under vacuum, and the crude material was purified by flash chromatography on silica gel to give the corresponding N-propargylic hydroxylamine.

General Procedure B for the Preparation of  $\beta$ -Enaminones (2a-t). In a 5-mL Wheaton reactor, equipped with a magnetic stirrer bar, the corresponding *N*-propargylic hydroxylamine (0.2 mmol, 1 equiv), and NaOH powder (0.02 mmol, 0.1 equiv) were added in acetonitrile (2 mL) by turn, and the mixture was heated at 50 °C for 1 h. Thus, the reaction mixture was filtered off through a silica gel pad and eluted with ethyl acetate, the filtrate was concentrated, and the residue

was purified by flash chromatography on silica gel to give the corresponding  $\beta$ -enaminone.

Gram-Scale Procedure for the Obtention of (Z)-Benzyl (3oxo-1-phenylhept-1-en-1-yl)carbamate (2a). In a 250 mL-roundbottom flask equipped with a magnetic stirrer bar and a condenser, the benzyl hydroxy(1-phenylhept-2-yn-1-yl)hept-2-yn-1-yl)carbamate 1a (1770 mg, 5.25 mmol, 1 equiv), NaOH powder (22 mg, 0.55 mmol, 0.1 equiv) were added in acetonitrile (50 mL), and the mixture was heated at 50 °C for 1 h. The reaction mixture was cooled and then filtered off through a silica gel pad and eluted with ethyl acetate, and the filtrate was concentrated. 1630 mg of the title compound 2a (4.83 mmol, 92% yield) and 70 mg of (*E*)-benzyl (1-phenylhept-2-yn-1-ylidene)carbamate 3a (0.21 mmol, 4% yield) were obtained after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 92/8).

General Procedure C for the Preparation of Pyrimidines (6, 9-22, 26, 27). In a 5 mL Wheaton reactor (screw top V-Vials with open-top cap) equipped with a magnetic stirrer bar, the corresponding vinylogous amide (0.2 mmol, 1 equiv), the corresponding amide (0.3 mmol, 1.5 equiv), 4 Å MS (70 mg) and potassium *tert*-butoxide (0.4 mmol, 2 equiv) were introduced by turn in toluene (2 mL), and the mixture was heated at 130 °C for 1 h. The reaction mixture was then filtered off through a silica gel pad and eluted with ethyl acetate. The filtrate was concentrated, and the residue was purified by flash chromatography on silica gel to give the corresponding pyrimidine.

Gram-Scale Procedure for the Obtention of 4-Butyl-2,6diphenylpyrimidine (6). In an oven-dried 100 mL-round-bottom flask equipped with a magnetic stirrer bar and a condenser, the (*Z*)benzyl (3-oxo-1-phenylhept-1-en-1-yl)carbamate 2a (1280 mg, 3.80 mmol, 1 equiv), benzamide (740 mg, 6.10 mmol, 1.6 equiv), 4 Å MS (1300 mg) and potassium *tert*-butoxide (908 mg, 8.mmol, 2.6 equiv) toluene were introduced by turn, and the mixture was heated at reflux for 1 h under argon. The reaction mixture was cooled and then filtered through a silica gel pad and eluted with ethyl acetate, and the filtrate was concentrated. 880 mg of the title compound 6 (3.05 mmol, 81% yield) were obtained after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 98/2).

General Procedure D for the Preparation of 2-Vinyl-Substituted Pyrimidines (23–25). In a 5 mL Wheaton reactor (screw top V-Vials with open-top cap) equipped with a magnetic stirrer bar, the corresponding vinylogous amide (0.2 mmol, 1 equiv), the corresponding amide (0.3 mmol, 1.5 equiv), 4 Å MS (70 mg), TEMPO (0.2 mmol, 1 equiv) and potassium *tert*-butoxide (0.4 mmol, 2 equiv) were introduced by turn in toluene (2 mL), and the mixture was heated at 130 °C for 1 h. The reaction mixture was then filtered off through a silica gel pad and eluted with ethyl acetate. The filtrate was concentrated, and the residue was purified by flash chromatography on silica gel to give the corresponding pyrimidine.

Benzyl hydroxy(1-phenylhept-2-yn-1-yl)carbamate (1a). According to general procedure A (on 2 mmol scale) and starting from 1phenylhept-2-yn-1-ol and benzyl hydroxycarbamate, 626.0 mg of compound 1a (1.855 mmol, 93% yield) were obtained as a pale yellow solid after purification by flash chromatography on silica gel (pentane/ ethyl acetate 88/12):  $R_f$  (silica, pentane/ethyl acetate 9/1 = 0.25, UV/ PMA); mp 58–59 °C (Et<sub>2</sub>O/pentane); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$ (ppm) 0.92 (t, J = 7.3 Hz, 3H), 1.39–1.59 (m, 4H), 2.29 (dt, J = 2.2 Hz and J = 7.1 Hz, 2H), 5.26 (s, 2H), 5.59 (brs, 1H), 6.13 (t, J = 2.1 Hz, 1H), 7.29-7.42 (m, 8H), 7.50-7.56 (m, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ (ppm) 13.6, 18.5, 22.0, 30.6, 55.9, 68.4, 74.5, 87.1, 127.9, 128.1, 128.2, 128.4, 128.6, 135.6, 136.2, 157.1; IR (ATR, neat)  $\nu$  (cm  $^{-1}$ ) 3225, 2957, 2928, 2872, 2859, 2222, 1959, 1704, 1495, 1481, 1453, 1443, 1394, 1346, 1290, 1097, 1030, 959, 763, 743, 713, 694, 636, 608; MS (ESI+) m/z 675 (15,  $[2M + H]^+$ ), 508 (45), 464 (100), 338 (50,  $[M + H]^+$ ), 320 (15), 261 (20); HRMS (ESI+) m/z 338.1742 calcd for C<sub>21</sub>H<sub>23</sub>NO<sub>3</sub> + H<sup>+</sup> 338.1756.

**Benzyl hydroxy(1-(***p***-tolyl)hept-2-yn-1-yl)carbamate (1b).** According to general procedure A (on 2 mmol scale) and starting from 1-*p*-tolylhept-2-yn-1-ol and benzyl hydroxycarbamate, 659.5 mg of compound **1b** (1.876 mmol, 94% yield) were obtained as a viscous pale yellow oil after purification by flash chromatography on silica gel (pentane/ethyl acetate 88/12):  $R_f$  (silica, pentane/ethyl acetate 9/1 = 0.25, UV/PMA); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 0.92 (t, *J* = 7.2 Hz, 3H), 1.38–1.58 (m, 4H), 2.28 (dt, *J* = 2.1 Hz and *J* = 7.0 Hz, 2H), 2.34 (s, 3H), 5.25 (s, 2H), 5.55 (brs, 1H), 6.09 (t, *J* = 2.1 Hz, 1H), 7.15 (~d, *J* = 8.0 Hz, 2H), 7.31–7.44 (m, 7H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 13.6, 18.5, 21.1, 22.0, 30.6, 55.6, 68.3, 74.7, 87.3, 127.9, 128.2, 128.4, 128.5, 129.1, 133.2, 135.6, 138.0, 157.1; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3314, 3032, 2956, 2931, 2871, 2233, 1702, 1513, 1497, 1455, 1409, 1350, 1328, 1282, 1214, 1180, 1093, 797, 771, 751, 734, 695; MS (ESI+) *m*/*z* 536 (15), 492 (15), 352 (10, [M + H]<sup>+</sup>), 334 (55), 216 (10), 185 (100); HRMS (ESI+) *m*/*z* 352.1910 calcd for C<sub>22</sub>H<sub>25</sub>NO<sub>3</sub> + H<sup>+</sup> 352 1913

Benzyl (1-(4-fluorophenyl)hept-2-yn-1-yl)(hydroxy)carbamate (1c). According to general procedure A (on 0.5 mmol scale) and starting from 1-(4-fluorophenyl)hept-2-yn-1-ol and benzyl hydroxycarbamate, 594.0 mg of compound 1c (1.671 mmol, 84% yield) were obtained as a pale yellow oil after purification by flash chromatography on silica gel (pentane/ethyl acetate 88/12):  $R_f$  (silica, pentane/ethyl acetate 9/1 = 0.25, UV/PMA); <sup>1</sup>H NMR (400 MHz,  $CDCl_3$ )  $\delta$  (ppm) 0.92 (t, J = 7.3 Hz, 3H), 1.39–1.58 (m, 4H), 2.28 (dt, J) = 2.1 Hz and J = 7.1 Hz, 2H), 5.23 (s, 2H), 6.03 (brs, 1H), 6.07 (brs, 1H), 7.01 (~t, J = 8.7 Hz, 2H), 7.31-7.40 (m, 5H), 7.49 (~dd, J = 5.4 Hz and J = 8.5 Hz, 2H). <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) –114.0; <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ (ppm) 13.6, 18.4, 22.0, 30.6, 55.1, 68.4, 74.5, 87.7, 115.2 (d,  $J_{ortho(C-F)} = 21.7 \text{ Hz}$ ), 128.2, 128.4, 128.6, 129.8 (d,  $J_{meta(C-F)} = 8.3 \text{ Hz}$ , 132.1 (d,  $J_{para(C-F)} = 3.0 \text{ Hz}$ ), 135.5, 157.1, 162.5 (d,  $J_{inso(C-F)} = 246.7 \text{ Hz}$ ; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3300, 2956, 2932, 2233, 1955, 1699, 1455, 1397, 1345, 1295, 1277, 1252, 1092, 1030, 797, 790, 777, 753, 734, 695, 634; MS (ESI+) m/z 523 (15), 356 (20,  $[M + H]^+$ ), 338 (100), 294 (25), 288 (10), 220 (40), 189 (90); HRMS (ESI+) m/z 356.1660 calcd for C<sub>21</sub>H<sub>22</sub>FNO<sub>3</sub> + H<sup>+</sup> 356.1662.

Benzyl (1-(4-chlorophenyl)hept-2-yn-1-yl)(hydroxy)carbamate (1d). According to general procedure A (on 4 mmol scale) and starting from 1-(4-chlorophenyl)hept-2-yn-1-ol and benzyl hydroxycarbamate, 1.290 g of compound 1d (3.469 mmol, 87% yield) were obtained as a viscous colorless oil after purification by flash chromatography on silica gel (pentane/ethyl acetate 88/12):  $R_f$  (silica, pentane/ethyl acetate 9/1 = 0.25, UV/PMA); <sup>1</sup>H NMR (400 MHz,  $CDCl_3$ )  $\delta$  (ppm) 0.92 (t, J = 7.3 Hz, 3H), 1.37–1.58 (m, 4H), 2.28 (dt, J = 2.1 Hz and J = 7.1 Hz, 2H), 5.23 (s, 2H), 5.97 (brs, 1H), 6.06 (brs, 1H), 7.30 (~d, J = 8.5 Hz, 2H), 7.32–7.39 (m, 5H), 7.45 (~d, J = 8.4 Hz, 2H);  ${}^{13}$ C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 13.6, 18.4, 22.0, 30.6, 55.2, 68.5, 74.2, 87.9, 128.2, 128.4, 128.5, 128.6, 129.4, 134.1, 134.9, 135.5, 157.2; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3297, 2929, 1706, 1491, 1455, 1407, 1292, 1091, 1016, 797, 749, 697; MS (ESI+) *m/z* 532 (10), 372 (20, [M + H]<sup>+</sup>), 295 (10), 205 (100), 189 (20); HRMS (ESI+) m/z 372.1377 calcd for C<sub>21</sub>H<sub>22</sub>ClNO<sub>3</sub> + H<sup>+</sup> 372.1366.

Benzyl (1-(4-bromophenyl)hept-2-yn-1-yl)(hydroxy)carbamate (1e). According to general procedure A (on 0.5 mmol scale) and starting from 1-(4-bromophenyl)hept-2-yn-1-ol and benzyl hydroxycarbamate, 177.9 mg of compound 1e (0.427 mmol, 86% yield) were obtained as a viscous colorless oil after purification by flash chromatography on silica gel (pentane/ethyl acetate 88/12):  $R_f$  (silica, pentane/ethyl acetate 9/1 = 0.25, UV/PMA); <sup>1</sup>H NMR (400 MHz,  $CDCl_3$ )  $\delta$  (ppm) 0.92 (t, J = 7.3 Hz, 3H), 1.37–1.57 (m, 4H), 2.28 (dt, J = 2.2 Hz and J = 7.1 Hz, 2H), 5.24 (s, 2H), 5.75 (brs, 1H), 6.04 (brs, 1H),7.31–7.41 (m, 7H), 7.46 (~d, J = 8.6 Hz, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ (ppm) 13.6, 18.4, 22.0, 30.6, 55.3, 68.5, 74.1, 88.0, 122.3, 128.2, 128.5, 128.6, 129.7, 131.5, 135.3, 135.4, 157.2; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3220, 2956, 2930, 2871, 2236, 1705, 1486, 1442, 1394, 1346, 1298, 1282, 1261, 1100, 1069, 1011, 954, 788, 745, 693, 672, 608; MS (ESI+) m/z 418 (90,  $[M + H]^{+,81}Br$ ), 416 (90,  $[M + H]^{+,79}Br$ ), 400 (98, <sup>81</sup>Br), 398 (100, <sup>79</sup>Br), 374 (35, <sup>81</sup>Br), 372 (35, <sup>79</sup>Br), 341 (30, <sup>81</sup>Br), 339 (30, <sup>79</sup>Br); HRMS (ESI+) m/z 416.0852 calcd for C<sub>21</sub>H<sub>22</sub>BrNO<sub>3</sub> + H<sup>+</sup> 416.0861.

Benzyl (1-([1,1'-biphenyl]-4-yl)hept-2-yn-1-yl)(hydroxy)carbamate (1f). According to general procedure A (on 1 mmol scale) and starting from 1-(biphenyl-4-yl)hept-2-yn-1-ol and benzyl hydroxycarbamate, 288.0 mg of compound 1f (0.696 mmol, 70% yield) were obtained as a viscous pale yellow oil after purification by flash chromatography on silica gel (pentane/ethyl acetate 9/1):  $R_f$  (silica, pentane/ethyl acetate 9/1 = 0.3, UV/PMA); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 0.99 (t, *J* = 7.2 Hz, 3H), 1.45–1.67 (m, 4H), 2.36 (dt, *J* = 2.0 Hz and *J* = 7.0 Hz, 2H), 5.26 (s, 2H), 6.23 (brs, 1H), 6.68 (brs, 1H), 7.31–7.52 (m, 8H), 7.58–7.70 (m, 6H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 13.5, 18.4, 21.9, 30.5, 55.4, 68.2, 74.7, 87.3, 126.9, 127.0, 127.3, 128.0, 128.2, 128.3, 128.4, 128.6, 135.4, 135.5, 140.5, 140.9, 157.2; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3300, 2956, 2932, 2871, 1955, 1699, 1455, 1397, 1345, 1295, 1277, 1092, 1030, 797, 790, 777, 734, 695, 634; MS (ESI+) *m*/*z* 414 (10, [M + H]<sup>+</sup>), 396 (25), 288 (15), 247 (100), 216 (10); HRMS (ESI+) *m*/*z* 414.2054 calcd for C<sub>27</sub>H<sub>27</sub>NO<sub>3</sub> + H<sup>+</sup> 414.2069.

Benzyl hydroxy(1-(m-tolyl)hept-2-yn-1-yl)carbamate (1g). According to general procedure A (on 1 mmol scale) and starting from 1-m-tolylhept-2-yn-1-ol and benzyl hydroxycarbamate, 291.3 mg of compound 1g (0.829 mmol, 83% yield) were obtained as a yellow oil after purification by flash chromatography on silica gel (pentane/ethyl acetate 88/12):  $R_f$  (silica, pentane/ethyl acetate 85/15 = 0.4, UV/ PMA); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 0.94 (t, *J* = 7.2 Hz, 3H), 1.39–1.60 (m, 4H), 2.30 (dt, J = 2.1 Hz and J = 7.0 Hz, 2H), 2.35 (s, 3H), 5.25 (s, 2H), 5.92 (brs, 1H), 6.10 (brs, 1H), 7.13 (~d, J = 7.3 Hz, 1H), 7.21-7.25 (m, 1H), 7.31-7.43 (m, 7H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ (ppm) 13.6, 18.5, 21.5, 22.0, 30.7, 55.8, 68.4, 74.8, 87.5, 125.1, 128.2, 128.3, 128.4, 128.6, 128.7, 129.0, 135.7, 136.3, 138.1, 157.2; IR  $(ATR, neat) \nu (cm^{-1}) 3302, 3033, 2956, 2931, 2871, 1702, 1455, 1404,$ 1350, 1281, 1092, 1029, 961, 753, 736, 719, 695; MS (ESI+) m/z 703 (5,  $[2M + H]^+$ , 536 (100), 519 (75), 492 (60), 459 (25), 369 (25), 352 (75, [M + H]<sup>+</sup>), 334 (80), 308 (60), 275 (75), 216 (25); HRMS (ESI+) m/z 352.1909 calcd for C<sub>22</sub>H<sub>25</sub>NO<sub>3</sub> + H<sup>+</sup> 352.1913.

Benzyl hydroxy(1-(3-methoxyphenyl)hept-2-yn-1-yl)carbamate (1h). According to general procedure A (on 2 mmol scale) and starting from 1-(3-methoxyphenyl)hept-2-yn-1-ol and benzyl hydroxycarbamate, 570.0 mg of compound **1h** (1.551 mmol, 78% yield) were obtained as an orange oil after purification by flash chromatography on silica gel (pentane/ethyl acetate 85/15): R<sub>f</sub> (silica, pentane/ ethyl acetate 85/15 = 0.3, UV/PMA); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$ (ppm) 0.93 (t, J = 7.3 Hz, 3H), 1.39–1.60 (m, 4H), 2.29 (dt, J = 2.2 Hz and J = 7.0 Hz, 2H), 3.76 (s, 3H), 5.23 (s, 2H), 6.11 (t, J = 2.1 Hz, 1H), 6.33 (brs, 1H), 6.84 (ddd, J = 0.7 Hz, J = 2.4 Hz and J = 8.1 Hz, 1H), 7.10–7.15 (m, 2H), 7.25 (t, J = 8.1 Hz, 1H), 7.30–7.41 (m, 5H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ (ppm) 13.5, 18.4, 21.9, 30.6, 55.1, 55.6, 68.3, 74.7, 87.3, 113.5, 113.8, 120.2, 128.1, 128.2, 128.5, 129.3, 135.6, 137.9, 157.1, 159.5; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3305, 2957, 2933, 2872, 1703, 1601, 1586, 1489, 1454, 1433, 1404, 1350, 1279, 1223, 1159, 1093, 1043, 755, 737, 719, 695, 639; MS (ESI+) m/z 568 (50), 401 (100), 368 (15, [M + H]<sup>+</sup>), 350 (20), 291 (80), 201 (50); HRMS (ESI +) m/z 368.1854 calcd for C<sub>22</sub>H<sub>25</sub>NO<sub>4</sub> + H<sup>+</sup> 368.1862.

Benzyl (1-(3-bromophenyl)hept-2-yn-1-yl)(hydroxy)carbamate (1i). According to general procedure A (on 2 mmol scale) and starting from 1-(3-bromophenyl)hept-2-yn-1-ol and benzyl hydroxycarbamate, 526.5 mg of compound 1i (1.264 mmol, 64% yield) were obtained as an colorless oil after purification by flash chromatography on silica gel (pentane/ethyl acetate  $9/1 \rightarrow 85/15$ ):  $R_f$  (silica, pentane/ethyl acetate 85/15 = 0.4, UV/PMA); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 0.93 (t, J = 7.3 Hz, 3H), 1.38–1.60 (m, 4H), 2.29 (dt, J = 2.2 Hz and J = 7.0 Hz, 2H), 5.26 (s, 2H), 5.67 (brs, 1H), 6.07 (t, J = 2.1 Hz, 1H), 7.19 (t, J = 7.9 Hz, 1H), 7.31-7.41 (m, 5H), 7.42-7.47 (m, 2H), 7.70 (t, J = 1.6 Hz, 1H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$ (ppm) 13.6, 18.4, 22.0, 30.6, 55.4, 68.6, 73.9, 88.3, 122.5, 126.6, 128.2, 128.5, 128.6, 129.9, 131.1, 131.3, 135.5, 138.6, 157.1; IR (ATR, neat) v  $(cm^{-1})\ 3291,\ 3065,\ 2956,\ 2932,\ 2871,\ 1703,\ 1593,\ 1571,\ 1455,\ 1423,$ 1350, 1281, 1093, 745, 730, 695, 673, 633; MS (ESI+) *m*/*z* 418 (30, [M  $(410, 79^{+81}), 416 (30, [M + H]^{+, 79}), 251 (98, 81^{-81}), 249 (100, 79^{-81});$ HRMS (ESI+) m/z 416.0852 calcd for C<sub>21</sub>H<sub>22</sub>BrNO<sub>3</sub> + H<sup>+</sup> 416.0861.

**Benzyl hydroxy(1-(thiophen-2-yl)hept-2-yn-1-yl)carbamate** (1j). According to general procedure A (on 3 mmol scale) and starting from 1-(thiophen-2-yl)hept-2-yn-1-ol and benzyl hydroxycarbamate, 886.0 mg of compound 1j (2.580 mmol, 86% yield) were obtained as a yellow oil after purification by flash chromatography on silica gel (pentane/ethyl acetate 9/1):  $R_f$  (silica, pentane/ethyl acetate 85/15 = 0.45, UV/PMA); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 0.92 (t, *J* = 7.2

Hz, 3H), 1.38–1.58 (m, 4H), 2.27 (dt, *J* = 2.2 Hz and *J* = 7.0 Hz, 2H), 5.24 (s, 2H), 6.00 (brs, 1H), 6.28 (brs, 1H), 6.95 (dd, *J* = 3.5 Hz and *J* = 5.1 Hz, 1H), 7.16 (td, *J* = 1.1 Hz and *J* = 3.5 Hz, 1H), 7.26 (td, *J* = 1.2 Hz and *J* = 5.2 Hz, 1H), 7.31–7.41 (m, 5H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 13.6, 18.4, 21.9, 30.5, 51.7, 68.5, 74.7, 86.7, 126.0, 126.6, 127.0, 128.2, 128.4, 128.5, 135.4, 139.8, 156.9; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3307, 2955, 2928, 2869, 1646, 1444, 1395, 1345, 1296, 1254, 1092, 932, 912, 820, 756, 695; MS (ESI+) *m*/*z* 344 (10, [M + H]<sup>+</sup>), 319 (25), 295 (15), 214 (15), 177 (100); HRMS (ESI+) *m*/*z* 344.1339 calcd for C<sub>19</sub>H<sub>21</sub>NO<sub>3</sub>S + H<sup>+</sup> 344.1320.

Benzyl hydroxy(1-(naphthalen-2-yl)hept-2-yn-1-yl)carbamate (1k). According to general procedure A (on 2 mmol scale) and starting from 1-(naphthalen-2-yl)hept-2-yn-1-ol and benzyl hydroxycarbamate, 706.0 mg of compound 1k (1.822 mmol, 91% yield) were obtained as an orange oil after purification by flash chromatography on silica gel (pentane/ethyl acetate  $9/1 \rightarrow 85/15$ ):  $R_f$  (silica, pentane/ethyl acetate 85/15 = 0.4, UV/PMA); <sup>1</sup>H NMR (400 MHz,  $CDCl_3$ )  $\delta$  (ppm) 0.95 (t, J = 7.3 Hz, 3H), 1.42–1.63 (m, 4H), 2.35 (dt, J = 2.2 Hz and J = 7.0 Hz, 2H), 5.28 (s, 2H), 5.62 (brs, 1H), 6.28 (brs, 1H), 7.31–7.44 (m, 5H), 7.46–7.52 (m, 2H), 7.59 (dd, J = 1.8 Hz and J = 8.5 Hz, 1H), 7.80-7.86 (m, 3H), 8.03 (brs, 1H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ (ppm) 13.6, 18.5, 22.0, 30.7, 56.0, 68.5, 74.6, 87.9, 125.5, 126.2, 126.3, 127.3, 127.6, 128.1, 128.2, 128.4, 128.6, 133.1, 133.2, 133.7, 135.6, 157.2; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3290, 3059, 2957, 2932, 2871, 1700, 1455, 1402, 1349, 1288, 1093, 808, 782, 739, 695; MS (ESI +) m/z 608 (90), 564 (50), 388 (10,  $[M + H]^+$ ), 370 (30), 311 (100); HRMS (ESI+) m/z 388.1919 calcd for C<sub>25</sub>H<sub>25</sub>NO<sub>3</sub> + H<sup>+</sup> 388.1913.

Benzyl hydroxy(1-(naphthalen-1-yl)hept-2-yn-1-yl)carbamate (11). According to general procedure A (on 1 mmol scale) and starting from 1-(naphthalen-2-yl)hept-2-yn-1-ol and benzyl hydroxycarbamate, 237.0 mg of compound 11 (0.613 mmol, 62% yield) were obtained as a brown oil after purification by flash chromatography on silica gel (pentane/ethyl acetate 9/1):  $R_f$  (silica, pentane/ethyl acetate 9/1 = 0.3, UV/PMA); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 0.95 (t, J = 7.3 Hz, 3H), 1.42-1.52 (m, 2H), 1.54-1.63 (m, 2H), 2.34 (dt, J = 2.0 Hz and J = 7.1 Hz, 2H), 5.23 (s, 2H), 5.73 (brs, 1H), 6.83 (brs, 1H), 7.32-7.43 (m, 5H), 7.44-7.52 (m, 3H), 7.81-7.87 (m, 2H), 8.02-8.08 (m, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 13.6, 18.5, 22.0, 30.6, 53.3, 68.3, 74.7, 88.0, 123.0, 125.0, 125.7, 126.5, 127.9, 128.1, 128.3, 128.5, 128.7, 129.3, 130.6, 131.2, 133.7, 135.7, 157.0; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3297, 2956, 2931, 2871, 1948, 1703, 1455, 1398, 1347, 1296, 1278, 1095, 797, 790, 778, 754, 737, 696; MS (ESI+) m/z 608 (10), 388 (5,  $[M + H]^+$ ), 311 (20), 221 (100); HRMS (ESI+) m/z388.1920 calcd for  $C_{25}H_{25}NO_3 + H^+$  388.1913.

Benzyl hydroxy(1-(o-tolyl)hept-2-yn-1-yl)carbamate (1m). According to general procedure A (on 1 mmol scale) and starting from 1-o-tolylhept-2-yn-1-ol and benzyl hydroxycarbamate, 243.5 mg of compound 1m (0.692 mmol, 70% yield) were obtained as a viscous orange oil after purification by flash chromatography on silica gel (pentane/ethyl acetate 88/12):  $R_f$  (silica, pentane/ethyl acetate 85/15 =0.4, UV/PMA); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 0.95 (t, J = 7.3 Hz, 3H), 1.40-1.60 (m, 4H), 2.30 (dt, J = 2.4 Hz and J = 7.2 Hz, 2H), 2.31 (s, 3H), 5.22 (s, 2H), 6.11 (brs, 1H), 6.22 (t, J = 2.0 Hz, 1H), 7.12-7.16 (m, 1H), 7.20-7.25 (m, 2H), 7.33-7.41 (m, 5H), 7.81-7.85 (m, 1H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ (ppm) 13.5, 18.4, 18.8, 21.9, 30.6, 53.3, 68.1, 74.9, 87.2, 125.7, 128.0, 128.2, 128.3, 128.4, 129.6, 130.2, 134.1, 135.7, 136.3, 157.1; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3263, 3034, 2960, 2868, 1675, 1655, 1487, 1463, 1455, 1410, 1346, 1296, 1259, 1093, 1079, 1016, 946, 797, 744, 729, 695, 629; MS (ESI+) m/z 536 (70), 519 (25), 492 (45), 459 (15), 369 (25), 352 (75, [M + H]<sup>+</sup>), 334 (20), 308 (50), 275 (100), 216 (30), 203 (30); HRMS (ESI+) m/z 352.1909 calcd for  $C_{22}H_{25}NO_3 + H^+$  352.1913.

**Benzyl (1-(2-fluorophenyl)hept-2-yn-1-yl)(hydroxy)carbamate (1n).** According to general procedure A (on 2 mmol scale) and starting from 1-(2-fluorophenyl)hept-2-yn-1-ol and benzyl hydroxycarbamate, 495.0 mg of compound **1n** (1.393 mmol, 70% yield) were obtained as a white semisolid after purification by flash chromatography on silica gel (pentane/ethyl acetate 9/1  $\rightarrow$  85/15):  $R_f$  (silica, pentane/ethyl acetate 85/15 = 0.4, UV/PMA); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 0.91 (t, J = 7.3 Hz, 3H, H<sub>1</sub>), 1.36–1.58 (m, 4H, H<sub>2</sub> and H<sub>3</sub>), 2.27 (dt, *J* = 2.1 Hz and *J* = 7.1 Hz, 2H, H<sub>4</sub>), 5.22 (brs, 2H, H<sub>15</sub>), 6.39 (brs, *J* = 2.1 Hz, 1H, H<sub>7</sub>), 6.99–7.06 (m, 1H, H<sub>Ar</sub>), 7.16 (dt, *J* = 1.0 Hz and *J* = 7.6 Hz, 1H, H<sub>Ar</sub>), 7.28–7.42 (m, 6H, H<sub>Ar</sub>), 7.78 (dt, *J* = 1.8 Hz and *J* = 7.6 Hz, 1H, H<sub>Ar</sub>), <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>) δ (ppm) -117.4; <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ (ppm) 13.5, 18.4, 21.9, 30.6, 49.8 (d, <sup>3</sup>*J*<sub>(C-F)</sub> = 4.3 Hz), 68.4, 74.0, 87.4, 115.3 (d, *J*<sub>ortho(C-F)</sub> = 21.3 Hz), 123.6 (d, <sup>2</sup>*J*<sub>(C-F)</sub> = 13.2 Hz), 123.8 (d, *J* = 3.6 Hz), 128.0, 128.2, 128.5, 130.2 (d, *J* = 8.4 Hz), 131.1 (d, *J*<sub>para(C-F)</sub> = 2.9 Hz), 135.7, 157.1, 160.1 (d, *J*<sub>ips0(C-F)</sub> = 249.4 Hz); IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3347, 2958, 2930, 2873, 1667, 1615, 1589, 1490, 1458, 1403, 1343, 1289, 1271, 1231, 1101, 1078, 952, 883, 753, 746, 698, 684, 626, 610; MS (ESI+) *m*/*z* 711 (35, [2M + H]<sup>+</sup>), 544 (25), 500 (90), 356 (50, [M + H]<sup>+</sup>), 189 (100); HRMS (ESI+) *m*/*z* 356.1645 calcd for C<sub>21</sub>H<sub>22</sub>FNO<sub>3</sub> + H<sup>+</sup> 356.1662.

Benzyl hydroxy(1-phenyl-3-(trimethylsilyl)prop-2-yn-1-yl)carbamate (10). According to general procedure A (on 4 mmol scale) and starting from 1-phenyl-3-(trimethylsilyl)prop-2-yn-1-ol and benzyl hydroxycarbamate, 1.230 g of compound 10 (3.479 mmol, 87% yield) were obtained as a colorless oil after purification by flash chromatography on silica gel (pentane/ethyl acetate 85/15):  $R_f$  (silica, pentane/ethyl acetate 85/15 = 0.3, UV/PMA); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 0.22 (s, 9H), 5.25 (s, 2H), 5.73 (brs, 1H), 6.15 (s, 1H), 7.30–7.42 (m, 8H), 7.50–7.55 (m, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) –0.1, 56.1, 68.5, 91.9, 99.7, 128.0, 128.2, 128.3, 128.4, 128.5, 128.6, 135.4, 135.5, 156.9; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3276, 3063, 3032, 2959, 1704, 1494, 1455, 1388, 1346, 1286, 1249, 1093, 1046, 1025, 838, 760, 751, 741, 693, 662, 604; MS (ESI+) *m*/*z* 707 (15, [2M + H]<sup>+</sup>), 540 (25), 354 (65, [M + H]<sup>+</sup>), 187 (100); HRMS (ESI+) *m*/*z* 354.1528 calcd for C<sub>20</sub>H<sub>23</sub>NO<sub>3</sub>Si + H<sup>+</sup> 354.1525.

**Benzyl hydroxy**(1-phenylbut-2-yn-1-yl)carbamate (1p). According to general procedure A (on 1 mmol scale) and starting from 1-phenylbut-2-yn-1-ol and benzyl hydroxycarbamate, 232.2 mg of compound 1p (0.786 mmol, 79% yield) were obtained as a pale yellow oil after purification by flash chromatography on silica gel (pentane/ethyl acetate 9/1  $\rightarrow$  88/12):  $R_f$  (silica, pentane/ethyl acetate 8/2 = 0.4, UV/PMA); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 1.92 (d, J = 2.4 Hz, 3H), 5.24 (s, 2H), 5.99 (brs, 1H), 6.10 (q, J = 2.0 Hz, 1H), 7.28–7.41 (m, 8H), 7.50 – 7.55 (m, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 3.7, 55.6, 68.4, 73.9, 82.9, 127.9, 128.1, 128.2, 128.4, 128.5, 135.6, 136.2, 157.1; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3297, 3065, 3033, 2921, 2251, 1955, 1699, 1496, 1452, 1403, 1350, 1283, 1093, 907, 727, 707, 695; MS (ESI +) m/z 591 (15, [2M + H]<sup>+</sup>), 424 (10), 380 (100), 296 (70, [M + H]<sup>+</sup>), 278 (90), 252 (15), 234 (50), 219 (40), 129 (20); HRMS (ESI+) m/z 296.1282 calcd for C<sub>18</sub>H<sub>17</sub>NO<sub>3</sub> + H<sup>+</sup> 296.1287.

Benzyl hydroxy(4-methyl-1-phenylpent-2-yn-1-yl)carbamate (1q). According to general procedure A (on 0.5 mmol scale) and starting from 4-methyl-1-phenylpent-2-yn-1-ol and benzyl hydroxycarbamate, 140.8 mg of compound 1q (0.436 mmol, 88% yield) were obtained as a pale yellow solid after purification by flash chromatography on silica gel (pentane/ethyl acetate  $9/1 \rightarrow 88/12$ ):  $R_f$  (silica, pentane/ethyl acetate 9/1 = 0.25, UV/PMA); mp 78-79 °C  $(Et_2O/pentane)$ ; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 1.21 (d, J = 6.9 Hz, 3H), 1.22 (d, J = 6.9 Hz, 3H), 2.66 (septd, J = 2.0 Hz and J = 6.8 Hz, 1H), 5.26 (s, 2H), 5.60 (brs, 1H), 6.13 (d, J = 1.8 Hz, 1H), 7.28–7.42 (m, 8H), 7.50–7.55 (m, 2H);  $^{13}$ C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 20.6, 22.9, 55.7, 68.4, 73.7, 93.0, 127.9, 128.1, 128.2, 128.4, 128.5, 135.6, 136.2, 157.1; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3300, 2967, 2927, 2877, 2255, 1967, 1888, 1754, 1655, 1495, 1448, 1415, 1349, 1288, 1099, 1030, 956, 883, 752, 739, 698, 634; MS (ESI+) m/z 647 (10,  $[2M + H]^+$ ), 491 (10), 480 (20), 436 (100), 324 (98, [M + H]<sup>+</sup>), 306 (25), 247 (25), 157 (20); HRMS (ESI+) m/z 324.1592 calcd for  $C_{20}H_{21}NO_3 + H^+$  324.1600.

Benzyl hydroxy(4,4-dimethyl-1-phenylpent-2-yn-1-yl)carbamate (1r). According to general procedure A (on 0.5 mmol scale) and starting from 4,4-dimethyl-1-phenylpent-2-yn-1-ol and benzyl hydroxycarbamate, 148.7 mg of compound 1r (0.441 mmol, 89% yield) were obtained as a pale yellow solid after purification by flash chromatography on silica gel (pentane/ethyl acetate 88/12):  $R_f$  (silica, pentane/ethyl acetate 9/1 = 0.25, UV/PMA); mp 100–102 °C (Et<sub>2</sub>O/ pentane); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 1.27 (s, 9H), 5.26 (s, 2H), 5.59 (brs, 1H), 6.12 (s, 1H), 7.28–7.42 (m, 8H), 7.50–7.54 (m, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 27.6, 30.9, 55.7, 68.3, 72.9, 95.9, 127.9, 128.1, 128.2, 128.4, 128.6, 135.6, 136.3, 157.1; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3268, 2965, 2925, 2250, 1950, 1798, 1697, 1455, 1388, 1348, 1291, 1267, 1213, 1093, 1026, 910, 825, 741, 761, 693, 676, 627; MS (ESI+) *m*/*z* 675 (5, [2M + H]<sup>+</sup>), 508 (35), 464 (20), 338 (100, [M + H]<sup>+</sup>), 320 (20), 261 (25), 171 (15); HRMS (ESI+) *m*/*z* 338.1741 calcd for C<sub>21</sub>H<sub>23</sub>NO<sub>3</sub> + H<sup>+</sup> 338.1756.

**Benzyl** (1,3-diphenylprop-2-yn-1-yl)(hydroxy)carbamate (1s). According to general procedure A (on 1 mmol scale) and starting from 1,3-diphenylprop-2-yn-1-ol and benzyl hydroxycarbamate, 259.7 mg of compound 1s (0.727 mmol, 73% yield) were obtained as a white solid after purification by flash chromatography on silica gel (pentane/ ethyl acetate 88/12):  $R_f$  (silica, pentane/ethyl acetate 9/1 = 0.3, UV/ PMA); mp 115–117 °C (Et<sub>2</sub>O/Pentane); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 5.28 (s, 2H), 5.73 (brs, 1H), 6.37 (s, 1H), 7.29–7.43 (m, 11H), 7.47–7.51 (m, 2H), 7.58–7.63 (m, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 56.1, 68.5, 83.9, 86.6, 122.2, 128.0, 128.2, 128.3, 128.4, 128.5, 128.5, 128.6, 128.7, 131.9, 135.5, 135.7, 157.1; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3298, 3063, 3033, 2949, 2233, 1955, 1702, 1490, 1454, 1443, 1406, 1351, 1285, 1268, 1098, 1028, 756, 707, 692; MS (ESI+) *m*/*z* 732 (15), 548 (10), 504 (10), 375 (20), 358 (20, [M + H]<sup>+</sup>), 296 (10), 214 (10), 191 (100), 171 (15), 102 (25); HRMS (ESI+) *m*/*z* 358.1432 calcd for C<sub>23</sub>H<sub>19</sub>NO<sub>3</sub> + H<sup>+</sup> 358.1443.

Benzyl hydroxy(1-phenyl-3-(*p*-tolyl)prop-2-yn-1-yl)carbamate (1t). According to general procedure A (on 4 mmol scale) and starting from 1-phenyl-3-(*p*-tolyl)prop-2-yn-1-ol and benzyl hydroxycarbamate, 1.080 g of compound 1t (2.907 mmol, 73% yield) were obtained as a yellow oil after purification by flash chromatography on silica gel (pentane/ethyl acetate 85/15):  $R_f$  (silica, pentane/ethyl acetate 85/15 = 0.3, UV/PMA); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 2.36 (s, 3H) 5.27 (s, 2H), 5.88 (brs, 1H), 6.36 (s, 1H), 7.13 (d, J = 7.6 Hz, 2H), 7.31–7.43 (m, 10H), 7.13 (d, J = 6.4 Hz, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 21.5, 56.1, 68.5, 83.2, 86.7, 119.1, 128.0, 128.2, 128.4, 128.5, 128.6, 129.0, 131.8, 135.6, 135.9, 138.8, 157.1; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3287, 2921, 1698, 1509, 1452, 1404, 1265, 1094, 816, 731, 695; MS (ESI+) *m*/*z* 576 (25), 532 (10), 372 (50, [M + H]<sup>+</sup>), 354 (35), 288 (25), 221 (15), 209 (30), 205 (100); HRMS (ESI+) *m*/*z* 372.1609 calcd for C<sub>24</sub>H<sub>21</sub>NO<sub>3</sub> + H<sup>+</sup> 372.1600.

**Benzyl hydroxy(4-phenylbut-3-yn-2-yl)carbamate (1v).** A 100 mL round-bottom flask charged with benzyl (*tert*-butyldimethylsilyl)oxycarbamate<sup>26</sup> (1.27 g, 4.52 mmol, 1.1 equiv), THF (20 mL), PPh<sub>3</sub> (1.19 g, 4.52 mmol, 1.1 equiv) and DIAD (1.02 mL, 5.15 mmol, 1.25 equiv) was cooled to 0 °C, and a solution of 4-phenylbut-3-yn-2-ol (603 mg, 4.12 mmol, 1 equiv) in THF (15 mL) was added dropwise. Then, the reaction was allowed to warm to ambient temperature and stirred for 2 h. Upon completion, the reaction was concentrated in vacuo and purified by flash chromatography on silica gel (Pentane/Et<sub>2</sub>O 98/2), which afforded 1.01 g of benzyl (*tert*-butyldimethylsilyl)oxy(4-phenylbut-3-yn-2-yl)carbamate with the unseparable DIAD excess (~20% by <sup>1</sup>H NMR). The product was used as such in the next step:  $R_f$  (silica, Pentane/Et<sub>2</sub>O 9/1 = 0.8, UV/PMA/KMnO<sub>4</sub>).

To a solution of benzyl (tert-butyldimethylsilyl)oxy(4-phenylbut-3yn-2-yl)carbamate (1.01 g, 2.47 mmol) in dry THF (12.5 mL) was added TBAF (1 M in THF, 2.5 mL, 2.5 mmol, 1.01 equiv) at 0 °C. After 5 min the mixture was quenched with saturated  $NH_4Cl$  solution (10 mL) and diluted with ether (20 mL). The organic layer was washed with water (20 mL) and brine (20 mL), dried over MgSO<sub>4</sub>, and evaporated, and the crude was further purified by column chromatography on silica (Pentane/Ethyl acetate 8/2) to give 619 mg of the title compound 1v (2.09 mmol, 51% yield over 2 steps) as a white solid after purification by flash chromatography on silica gel (pentane/ethyl acetate 85/15):  $R_f$ (silica, pentane/ethyl acetate 8/2 = 0.3, UV/PMA); mp 90-91 °C  $(Et_2O/pentane)$ ; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 1.57 (d, J = 7.2 Hz, 3H, H<sub>1</sub>), 5.15–5.25 (m, 3H, H<sub>2</sub> and H<sub>11</sub>), 6.82 (brs, 1H, H<sub>3</sub>), 7.25– 7.44 (m, 10H,  $H_{Ar}$ ); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 18.8, 48.2, 68.3, 83.5, 86.7, 122.4, 128.1, 128.0, 128.1, 128.2, 128.3, 128.5, 131.8, 135.6, 157.2; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3279, 2985, 2939, 1954, 1698, 1489, 1454, 1443, 1402, 1353, 1289, 1114, 1068, 753, 690; MS (ESI+) m/z 591 (30,  $[2M + H]^+$ ), 419 (25), 296 (100,  $[M + H]^+$ ), 252 (25), 234 (40), 193 (40); HRMS (ESI+) m/z 296.1276 calcd for C<sub>18</sub>H<sub>17</sub>NO<sub>3</sub> + H<sup>+</sup> 296.1287.

Benzyl hydroxy(2-methyldec-5-yn-4-yl)carbamate (1w). An anhydrous and purged with N<sub>2</sub> 10 mL round-bottom flask equipped with a magnetic stirrer bar was charged with dry THF (1 mL) and 1hexyne (0.23 mL, 2 mmol, 4 equiv), and the resulting solution was cooled at -78 °C. n-Butyllithium (1.6 M in hexanes, 1.25 mL, 2 mmol, 4 equiv) was then added dropwise, and after 10 min, a solution of benzyl hydroxy(3-methyl-1-(phenylsulfonyl)butyl)carbamate (189 mg, 0.5 mmol, 1 equiv) in dry THF (1 mL) was added via syringe. The resulting mixture was stirred 30 min at -78 °C and then allowed to run at 0 °C for 2 h. A saturated aqueous NH4Cl (2 mL) was then added, and the solution was allowed to reach room temperature. The reaction mixture was then extracted with ethyl acetate (2  $\times$  10 mL), and the combined organic layers were washed with brine (5 mL), dried over anhydrous MgSO4 and filtered. After the removal of the solvent, the crude mixture was purified by column chromatography (Pentane/Ethyl acetate 95/5) to give 4 mg of the title compound 1w (0.013 mmol, 3% yield) as a white solid after purification by flash chromatography on silica gel (pentane/ethyl acetate 95/5):  $R_f$  (silica, pentane/ethyl acetate 85/15 = 0.7, UV/PMA); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 0.86–0.94 (m, 9H), 1.32–1.77 (m, 7H), 2.17 (dt, J = 2.0 Hz and J = 7.0 Hz, 2H), 4.87 (tt, J = 2.0 Hz and J = 7.6 Hz, 1H), 5.22 (s, 2H), 5.59 (brs, 1H), 7.30-7.42 (m, 5H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 13.6, 18.3, 21.9, 22.2, 22.3, 24.8, 30.7, 41.5, 50.9, 68.2, 73.4, 85.2, 128.1, 128.3, 128.5, 135.7, 157.3; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3254, 2959, 2932, 2871, 1970, 1704, 1456, 1305, 1260, 1091, 1021, 798; MS (ESI+) m/z 635 (25, [2M + H]<sup>+</sup>), 608 (25), 410 (25), 359 (15), 324 (15), 318 (100, [M + H]<sup>+</sup>), 288 (65), 241 (20), 210 (35), 209 (25); HRMS (ESI+) m/z 318.2064 calcd for C<sub>19</sub>H<sub>27</sub>NO<sub>3</sub> + H<sup>+</sup> 318.2069.

(Z)-Benzyl (3-oxo-1-phenylhept-1-en-1-yl)carbamate (2a). According to general procedure B (on 0.2 mmol scale) and starting from benzyl hydroxy(1-phenylhept-2-yn-1-yl)hept-2-yn-1-yl)carbamate 1a, 57.3 mg of compound 2a (0.170 mmol, 86% yield) were obtained as a pale yellow oil after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 92/8):  $R_f$  (silica, pentane/Et<sub>2</sub>O 9/1 = 0.35, UV/ Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 0.92 (t, J = 7.2 Hz, 3H), 1.29–1.41 (m, 2H), 1.55–1.67 (m, 2H), 2.47 (t, J = 7.0 Hz, 2H), 5.07 (s, 2H), 5.57 (s, 1H), 7.28-7.46 (m, 10H), 11.58 (brs, 1H, NH); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ (ppm) 13.9, 22.4, 26.9, 43.5, 67.4, 107.2, 127.5, 128.0, 128.3, 128.4, 128.5, 129.8, 135.5, 135.6, 152.8, 154.5, 202.6; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 2958, 2928, 2870, 1750, 1642, 1589, 1573, 1495, 1473, 1280, 1195, 1119, 1047, 765, 750, 694; MS (ESI+) *m*/  $z 675 (5, [2M + H]^+), 505 (10), 338 (100, [M + H]^+), 320 (15), 294$ (10); HRMS (ESI+) m/z 338.1747 calcd for  $C_{21}H_{23}NO_3 + H^+$ 338.1756.

(*E*)-Benzyl (1-phenylhept-2-yn-1-ylidene)carbamate (3a). According to the gram-scale procedure for the preparation of (*Z*)-benzyl (3-oxo-1-phenylhept-1-en-1-yl)carbamate 2a, 70 mg of compound 3a (0.21 mmol, 4% yield) were isolated as a minor product:  $R_f$  (silica, Pentane/Et<sub>2</sub>O 9/1 = 0.45, UV/PMA); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 0.93 (t, *J*= 7.2 Hz, 3H), 1.38–1.60 (m, 4H), 2.36 (t, *J*= 7.2 Hz, 2H), 5.33 (s, 2H), 7.30–7.55 (m, 8H), 8.08 (~d, *J* = 7.2 Hz, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 13.5, 19.0, 21.9, 29.8, 68.2, 74.1, 103.4, 128.3, 128.4, 128.5, 132.4, 135.0, 135.6, 153.6, 163.1; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3066, 3033, 2972, 2929, 2868, 2218, 1724, 1597, 1571, 1450, 1318, 1295, 1258, 1203, 1176, 906, 690; MS (ESI+) *m/z* 320 (100, [M + H]<sup>+</sup>), 276 (30), 187 (15); HRMS (ESI+) *m/z* 320.1649 calcd for C<sub>21</sub>H<sub>21</sub>NO<sub>2</sub> + H<sup>+</sup> 320.1651.

(Z)-Benzyl (3-oxo-1-(*p*-tolyl)hept-1-en-1-yl)carbamate (2b). According to general procedure B (on 0.1 mmol scale) and starting from benzyl (1-(*p*-tolyl)hept-2-yn-1-yl)(hydroxy)carbamate 1b, 27.7 mg of compound 2b (0.0788 mmol, 90% yield) were obtained as a pale yellow oil after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 92/8):  $R_f$  (silica, pentane/Et<sub>2</sub>O 9/1 = 0.35, UV/ Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 0.92 (t, *J* = 7.2 Hz, 3H), 1.30–1.41 (m, 2H), 1.56–1.66 (m, 2H), 2.38 (s, 3H), 2.46 (t, *J* = 7.0 Hz, 2H), 5.08 (s, 2H), 5.57 (s, 1H), 7.19 (~d, *J* = 8 Hz, 2H), 7.28– 7.37 (m, 7H), 11.58 (brs, 1H, NH); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$ (ppm) 13.9, 21.4, 22.4, 26.8, 43.4, 67.3, 106.8, 127.4, 128.2, 128.4, 128.7, 132.6, 135.5, 140.0, 152.8, 154.5, 202.4; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3032, 2957, 2930, 2871, 1956, 1905, 1750, 1641, 1589, 1567, 1474, 1455, 1286, 1195, 1136, 1047, 810, 764, 754, 696; MS (ESI+) m/z 703 (5, [2M + H]<sup>+</sup>), 352 (100, [M + H]<sup>+</sup>), 334 (55), 308 (10), 210 (10), 154 (10); HRMS (ESI+) m/z 352.1913 calcd for C<sub>22</sub>H<sub>25</sub>NO<sub>3</sub> + H<sup>+</sup> 352.1913.

(Z)-Benzyl (1-(4-fluorophenyl)-3-oxohept-1-en-1-yl)carbamate (2c). According to general procedure B (on 0.2 mmol scale) and starting from benzyl (1-(4-fluorophenyl)hept-2-yn-1-yl)-(hydroxy)carbamate 1c, 48.0 mg of compound 2c (0.135 mmol, 71% yield) were obtained as a red oil after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 92/8): R<sub>f</sub> (silica, pentane/Et<sub>2</sub>O 9/1 = 0.4, UV/Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 0.93 (t, J = 7.2 Hz, 3H), 1.31–1.41 (m, 2H), 1.58–1.66 (m, 2H), 2.47 (t, J = 7.6 Hz, 2H), 5.08 (s, 2H), 5.54 (s, 1H), 7.19 (~t, J = 8.4 Hz, 2H), 7.29-7.42 (m, 7H), 11.58 (brs, 1H, NH). <sup>19</sup>F NMR (376 MHz,  $CDCl_3$ )  $\delta$  (ppm) –110.6; <sup>13</sup>C NMR (100 MHz,  $CDCl_3$ )  $\delta$  (ppm) 13.9, 22.4, 26.8, 43.5, 67.3, 107.2, 115.2 (d, *J*<sub>ortho(C-F)</sub> = 22.0 Hz), 128.3, 128.4, 128.6, 129.5 (d,  $J_{meta(C-F)} = 8.4 \text{ Hz}$ ), 131.5 (d,  $J_{para(C-F)} = 3.4 \text{ Hz}$ ), 135.4, 152.9, 153.3, 163.6 (d,  $J_{ipso(C-F)} = 249.9$  Hz), 202.5; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 2958, 2930, 2872, 1748, 1644, 1604, 1582, 1507, 1473, 1455, 1286, 1192, 1157, 1134, 1074, 1044, 839, 763, 731, 695; MS (ESI +) m/z 356 (100,  $[M + H]^+$ ), 338 (15), 312 (20); HRMS (ESI+) m/z356.1669 calcd for C<sub>21</sub>H<sub>22</sub>FNO<sub>3</sub> + H<sup>+</sup> 356.1662.

(Z)-Benzyl (1-(4-chlorophenyl)-3-oxohept-1-en-1-yl)carbamate (2d). According to general procedure B (on 0.1 mmol scale) and starting from benzyl (1-(4-chlorophenyl)hept-2-yn-1-yl)-(hydroxy)carbamate 1d, 29.5 mg of compound 2d (0.794 mmol, 80% yield) were obtained as a yellow oil after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 92/8): R<sub>f</sub> (silica, pentane/Et<sub>2</sub>O 9/1 = 0.4, UV/Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 0.92 (t, J = 7.3 Hz, 3H), 1.30–1.40 (m, 2H), 1.56–1.66 (m, 2H), 2.47 (t, J = 7.2 Hz, 2H), 5.07 (s, 2H), 5.54 (s, 1H), 7.28-7.38 (m, 9H), 11.54 (brs, 1H, NH);  $^{13}$ C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 13.9, 22.4, 26.7, 43.5, 67.5, 107.2, 128.3, 128.4, 128.5, 128.8, 134.0, 135.3, 135.8, 152.8, 153.0, 202.5; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3066, 3034, 2957, 2930, 2872, 1748, 1644, 1598, 1585, 1490, 1471, 1281, 1193, 1135, 1073, 1045, 1013, 908, 815, 729, 695; MS (ESI+) m/z 743 (5, [2M + H]<sup>+</sup>), 372 (100,  $[M + H]^+$ ), 328 (15); HRMS (ESI+) m/z 372.1364 calcd for C<sub>21</sub>H<sub>22</sub>ClNO<sub>3</sub> + H<sup>+</sup> 372.1366.

(Z)-Benzyl (1-(4-bromophenyl)-3-oxohept-1-en-1-yl)carbamate (2e). According to general procedure B (on 0.1 mmol scale) and starting from benzyl (1-(4-bromophenyl)hept-2-yn-1yl)(hydroxy)carbamate 1e, 36.1 mg of compound 2e (0.867 mmol, 87% yield) were obtained as a red oil after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 92/8): R<sub>f</sub> (silica, Pentane/ Et<sub>2</sub>O 9/1 = 0.4, UV/Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>2</sub>)  $\delta$  (ppm) 0.92 (t, J = 7.3 Hz, 3H), 1.29-1.40 (m, 2H), 1.54-1.65 (m, 2H), 2.47 (t, J = 7.6 Hz, 2H), 5.07 (s, 2H), 5.53 (s, 1H), 7.26 (~d, J = 8.4 Hz, 2H), 7.28–7.37 (m, 5H), 7.50 (~d, J = 8.4 Hz, 2H), 11.53 (brs, 1H, NH); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ (ppm) 13.9, 22.4, 26.8, 43.5, 67.6, 107.3, 124.2, 128.3, 128.4, 128.6, 129.1, 131.3, 134.5, 135.3, 152.8, 153.1, 202.5; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3020, 2958, 2930, 2872, 1746, 1596, 1582, 1561, 1489, 1472, 1281, 1194, 1137, 1067, 1047, 1010, 746, 695, 666; MS (ESI+) m/z 418 (97,  $[M + H]^{+,81}Br$ ), 416 (100, [M +H]<sup>+,79</sup>Br); HRMS (ESI+) m/z 416.0844 calcd for C<sub>21</sub>H<sub>22</sub>BrNO<sub>3</sub> + H<sup>+</sup> 416.0861.

(*Z*)-Benzyl (1-([1,1'-biphenyl]-4-yl)-3-oxohept-1-en-1-yl)carbamate (2f). According to general procedure B (on 0.2 mmol scale) and starting from benzyl (1-([1,1'-biphenyl]-4-yl)hept-2-yn-1yl)(hydroxy)carbamate 1f, 49.8 mg of compound 2f (0.120 mmol, 61% yield) were obtained as a pale yellow oil after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 92/8):  $R_f$  (silica, pentane/ Et<sub>2</sub>O 9/1 = 0.4, UV/Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 0.95 (t, *J* = 7.2 Hz, 3H), 1.32–1.44 (m, 2H), 1.58–1.70 (m, 2H), 2.50 (t, *J* = 7.2 Hz, 2H), 5.11 (s, 2H), 5.65 (s, 1H), 7.30–7.41 (m, 6H), 7.44– 7.52 (m, 4H), 7.60–7.65 (m, 4H), 11.62 (brs, 1H, NH); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 13.9, 22.4, 26.8, 43.4, 67.4, 107.2, 126.7, 127.1, 127.6, 128.0, 128.2, 128.3, 128.5, 128.8, 134.4, 135.4, 140.3, 142.6, 152.9, 154.0, 202.4; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3062, 3032, 2956, 2930, 2871, 1748, 1587, 1578, 1472, 1280, 1192, 1134, 1075, 1047, 908, 763, 752, 730, 694; MS (ESI+) *m*/*z* 414 (100, [M + H]<sup>+</sup>), 288 (10), 252 (10), 216 (25), 203 (5); HRMS (ESI+) m/z 414.2065 calcd for C<sub>27</sub>H<sub>27</sub>NO<sub>3</sub> + H<sup>+</sup> 414.2069.

(Z)-Benzyl (3-oxo-1-(*m*-tolyl)hept-1-en-1-yl)carbamate (2g). According to general procedure B (on 0.4 mmol scale) and starting from benzyl (1-(*m*-tolyl)hept-2-yn-1-yl)(hydroxy)carbamate 1g, 103.5 mg of compound 2g (0.295 mmol, 74% yield) were obtained as a yellow oil after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 92/8):  $R_f$  (silica, pentane/Et<sub>2</sub>O 9/1 = 0.4, UV/Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 0.91 (t, J = 7.3 Hz, 3H), 1.29–1.39 (m, 2H), 1.56–1.65 (m, 2H), 2.36 (s, 3H), 2.45 (t, J = 7.2 Hz, 2H), 5.06 (s, 2H), 5.55 (s, 1H), 7.16–7.36 (m, 9H), 11.56 (brs, 1H, NH); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 13.8, 21.3, 22.3, 26.7, 43.3, 67.3, 107.0, 124.6, 127.8, 127.9, 128.1, 128.2, 128.4, 130.5, 135.4, 135.5, 137.6, 152.7, 154.5, 202.4; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3033, 2957, 2929, 2871, 1750, 1642, 1576, 1472, 1455, 1288, 1204, 1188, 1135, 1049, 786, 763, 754, 733, 695; MS (ESI+) *m*/*z* 352 (100, [M + H]<sup>+</sup>), 308 (10); HRMS (ESI+) *m*/*z* 352.1929 calcd for C<sub>22</sub>H<sub>25</sub>NO<sub>3</sub> + H<sup>+</sup> 352.1913.

(Z)-Benzyl (1-(3-methoxyphenyl)-3-oxohept-1-en-1-yl)carbamate (2h). According to general procedure B (on 0.4 mmol scale) and starting from benzyl (1-(3-methoxyphenyl)hept-2-yn-1yl)(hydroxy)carbamate 1h, 96.5 mg of compound 2h (0.263 mmol, 66% yield) were obtained as a pale yellow oil after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 92/8):  $R_f$  (silica, pentane/  $Et_2O 9/1 = 0.3$ , UV/Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 0.94 (t, J = 7.2 Hz, 3H), 1.32–1.43 (m, 2H), 1.58–1.68 (m, 2H), 2.48 (t, J = 7.2 Hz, 2H), 3.81 (s, 3H), 5.09 (s, 2H), 5.60 (s, 1H), 6.92-7.46 (m, 3H), 7.28-7.39 (m, 6H), 11.55 (brs, 1H, NH). <sup>19</sup>F NMR (376 MHz,  $CDCl_3$ )  $\delta$  (ppm) -114.1; <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 13.8, 22.3, 26.7, 43.4, 55.2, 67.3, 107.0, 113.1, 115.1, 119.8, 128.2, 128.4, 128.9, 135.4, 136.9, 152.6, 154.0, 159.1, 202.4; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3066, 3033, 2957, 2932, 1749, 1644, 1574, 1464, 1287, 1223, 1191, 1134, 1039, 880, 785, 755, 733, 695; MS (ESI+) *m/z* 368 (100, M +  $H^{+}$ , 324 (20); HRMS (ESI+) m/z 368.1855 calcd for  $C_{22}H_{25}NO_4$  + H<sup>+</sup> 368.1862.

(Z)-Benzyl (1-(3-bromophenyl)-3-oxohept-1-en-1-yl)carbamate (2i). According to general procedure B (on 0.4 mmol scale) and starting from benzyl (1-(3-bromophenyl)hept-2-yn-1yl)(hydroxy)carbamate 1i, 86.0 mg of compound 2i (0.206 mmol, 52% yield) were obtained as a pale yellow oil after purification by flash chromatography on silica gel (petroleum ether/Et<sub>2</sub>O 92/8):  $R_f$  (silica, pentane/Et<sub>2</sub>O 9/1 = 0.4, UV/Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$ (ppm) 0.93 (t, J = 7.3 Hz, 3H), 1.31-1.42 (m, 2H), 1.57-1.68 (m, 2H),2.48 (t, J = 7.2 Hz, 2H), 5.08 (s, 2H), 5.54 (s, 1H), 7.21–7.27 (m, 1H), 7.28-7.49 (m, 6H), 7.52-7.56 (m, 2H), 11.51 (brs, 1H, NH); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ (ppm) 13.8, 22.3, 26.6, 43.4, 67.5, 107.4, 122.0, 126.2, 128.2, 128.3, 128.4, 129.3, 130.2, 132.5, 135.3, 137.6, 152.4, 152.6, 202.4; IR (ATR, neat) ν (cm<sup>-1</sup>) 3065, 3034, 2957, 2930, 2871, 1748, 1582, 1600, 1558, 1479, 1463, 1280, 1192, 1131, 1046, 909, 785, 729, 694; MS (ESI+) m/z 418 (97,  $[M + H]^{+,81}Br$ ), 416 (100, [M +H]<sup>+,79</sup>Br), 148 (45); HRMS (ESI+) m/z 416.0860 calcd for  $C_{21}H_{22}BrNO_3 + H^+ 416.0861.$ 

(Z)-Benzyl (3-oxo-1-(thiophen-2-yl)hept-1-en-1-yl)carbamate (2j). According to general procedure B (on 1 mmol scale) and starting from benzyl hydroxy(1-(thiophen-2-yl)hept-2-yn-1yl)carbamate 1j, 155.0 mg of compound 2j (0.451 mmol, 45% yield) were obtained as a red oil after purification by flash chromatography on silica gel (petroleum ether/Et<sub>2</sub>O 92/8):  $R_f$  (silica, Pentane/Et<sub>2</sub>O 9/1 = 0.4, UV/Vanillin); <sup>1</sup>H NMR (400 MHz,  $CDCl_3$ )  $\delta$  (ppm) 0.94 (t, J = 7.3 Hz, 3H), 1.32–1.42 (m, 2H), 1.58–1.67 (m, 2H), 2.48 (t, J = 7.6 Hz, 2H), 5.13 (s, 2H), 5.78 (s, 1H), 7.06 (dd, J = 3.7 Hz and J = 5.0 Hz, 1H), 7.33 (td, J = 1.2 Hz and J = 3.7 Hz, 1H), 7.34–7.38 (m, 5H), 7.42 (dd, J =1.2 Hz and J = 5.0 Hz, 1H), 11.45 (brs, 1H, NH); <sup>13</sup>C NMR (100 MHz,  $CDCl_3$   $\delta$  (ppm) 13.8, 22.3, 26.7, 43.4, 67.4, 107.2, 127.2, 128.0, 128.2, 128.3, 128.4, 129.2, 135.4, 136.5, 146.9, 152.8, 202.2; IR (ATR, neat)  $\nu$ (cm<sup>-1</sup>) 2957, 2930, 1749, 1639, 1582, 1464, 1455, 1191, 1267, 1191, 1134, 1109, 694; MS (ESI+) m/z 344 (100,  $[M + H]^+$ ), 300 (30); HRMS (ESI+) m/z 344.1313 calcd for  $C_{19}H_{21}NO_3S + H^+$  344.1320.

(Z)-Benzyl (1-(naphthalen-2-yl)-3-oxohept-1-en-1-yl)carbamate (2k). According to general procedure B (on 0.4 mmol scale) and starting from benzyl (1-(naphthalen-2-yl)hept-2-yn-1-

yl)(hydroxy)carbamate 1k, 87.1 mg of compound 2k (0.225 mmol, 58% yield) were obtained as a pale yellow oil after purification by flash chromatography on silica gel (petroleum ether/Et<sub>2</sub>O 92/8):  $R_f$  (silica, Pentane/Et<sub>2</sub>O 9/1 = 0.4, UV/PMA/Vanillin); <sup>1</sup>H NMR (400 MHz,  $CDCl_3$ )  $\delta$  (ppm) 0.98 (t, J = 7.6 Hz, 3H, H<sub>1</sub>), 1.36–1.47 (m, 2H, H<sub>2</sub>), 1.64–1.73 (m, 2H, H<sub>3</sub>), 2.53 (t, J = 7.2 Hz, 2H, H<sub>4</sub>), 5.11 (s, 2H, H<sub>18</sub>), 5.72 (s, 1H, H<sub>6</sub>), 7.28–7.36 (m, 5H, H<sub>Ar</sub>), 7.48–7.58 (m, 3H, H<sub>Ar</sub>), 7.81-7.95 (m, 4H, H<sub>Ar</sub>), 11.71 (brs, 1H, NH); <sup>13</sup>C NMR (100 MHz,  $CDCl_3$ )  $\delta$  (ppm) 13.8 (C<sub>1</sub>), 22.3 (C<sub>2</sub>), 26.7 (C<sub>3</sub>), 43.4 (C<sub>4</sub>), 67.4 (C<sub>18</sub>), 107.4 (C<sub>6</sub>), 125.1 (C<sub>Ar</sub>), 126.4 (C<sub>Ar</sub>), 126.7 (C<sub>Ar</sub>), 126.9 (C<sub>Ar</sub>), 127.3  $(C_{Ar})$ , 127.7  $(C_{Ar})$ , 128.1  $(C_{Ar})$ , 128.2  $(C_{Ar})$ , 128.3  $(2C, C_{Ar})$ , 128.4  $(2C, C_{Ar})$  $C_{Ar}$ ), 132.7 ( $C_{qr}$ ,  $C_{Ar}$ ), 133.2 ( $C_{qr}$ ,  $C_{Ar}$ ), 133.8 ( $C_{qr}$ ,  $C_{Ar}$ ), 135.4 ( $C_{qr}$ ,  $C_{Ar}$ ), 152.8 ( $C_{qr}$ ,  $C_{Ar}$ ), 154.3 ( $C_{qr}$ ,  $C_{Ar}$ ), 202.4 ( $C_{5}$ ); IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3060, 2958, 2931, 2872, 1747, 1641, 1584, 1478, 1463, 1361, 1284, 1191, 1135, 1109, 1047, 906, 814, 726, 695, 646; MS (ESI+) m/z 388  $(100, [M + H]^+)$ , 370 (20), 148 (20); HRMS (ESI+) m/z 388.1920 calcd for C<sub>25</sub>H<sub>25</sub>NO<sub>3</sub> + H<sup>+</sup> 388.1913.

(Z)-Benzyl (1-(naphthalen-1-yl)-3-oxohept-1-en-1-yl)carbamate (21). According to general procedure B (on 0.2 mmol scale) and starting from benzyl (1-(naphthalen-1-yl)hept-2-yn-1yl)(hydroxy)carbamate 11, 18.5 mg of compound 21 (0.048 mmol, 29% yield) were obtained as a pale yellow oil after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 92/8): R<sub>f</sub> (silica, Pentane/ Et<sub>2</sub>O 9/1 = 0.4, UV/PMA/Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$ (ppm) 0.93 (t, J = 7.2 Hz, 3H), 1.32–1.44 (m, 2H), 1.60–1.68 (m, 2H), 2.48 (t, J = 7.2 Hz, 2H), 4.92 (s, 2H), 5.59 (s, 1H), 7.15-7.21 (m, 2H), 7.27-7.32 (m, 3H), 7.40 (dd, J = 1.1 Hz and J = 7.1 Hz, 1H), 7.45-7.54 (m, 3H), 7.81-7.94 (m, 3H), 12.05 (brs, 1H, NH); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 13.9, 22.4, 26.8, 43.4, 67.3, 107.0, 124.4, 124.9, 125.3, 126.0, 126.7, 128.1, 128.2, 128.4, 128.5, 129.4, 131.0, 132.9, 133.7, 135.3, 152.0, 153.4, 202.9; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3062, 2957, 2930, 2871, 1754, 1642, 1591, 1575, 1477, 1286, 1200, 1179, 1140, 1092, 801, 776, 697; MS (ESI+) m/z 388 (100,  $[M + H]^+$ ), 344 (10), 320 (15), 288 (25), 214 (10), 187 (10); HRMS (ESI+) m/z 388.1920 calcd for  $C_{25}H_{25}NO_{2} + H^{+}$  388.1913.

(Z)-Benzyl (1-(2-fluorophenyl)-3-oxohept-1-en-1-yl)carbamate (2n). According to general procedure B (on 0.4 mmol scale) and starting from benzyl (1-(2-fluorophenyl)hept-2-yn-1-yl)-(hydroxy)carbamate 1n, 119.9 mg of compound 2n (0.337 mmol, 84% yield) were obtained as a yellow oil after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 92/8): R<sub>f</sub> (silica, Pentane/Et<sub>2</sub>O 9/1 = 0.4, UV/PMA/Vanillin); <sup>1</sup>H NMR (400 MHz,  $CDCl_3$ )  $\delta$  (ppm) 0.93 (t, J = 7.3 Hz, 3H), 1.31–1.42 (m, 2H), 1.57–1.67 (m, 2H), 2.47 (t, J = 7.2 Hz, 2H), 5.09 (s, 2H), 5.50 (s, 1H), 7.08 (~t, J =9.6 Hz, 1H), 7.18 (~t, J = 7.5 Hz, 1H), 7.27-7.46 (m, 7H), 11.77 (brs, 1H, NH). <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) –114.1; <sup>13</sup>C NMR  $(100 \text{ MHz}, \text{CDCl}_3) \delta$  (ppm) 13.8, 22.3, 26.7,  $\overline{43.4}$ , 67.4, 106.9, 115.1 (d,  $J_{ortho(C-F)} = 22.0 \text{ Hz}$ , 123.7 (d,  $J_{meta(C-F)} = 14.7 \text{ Hz}$ ), 123.9 (d,  $J_{para(C-F)} = 14.7 \text{ Hz}$ ) 3.5 Hz), 128.1, 128.2, 128.4, 129.3 (d,  $J_{(C-F)} = 2.9$  Hz), 131.2 (d,  $J_{meta(C-F)} = 8.3 \text{ Hz}$ , 135.4, 149.0, 152.4, 159.7 (d,  $J_{ipso(C-F)} = 249.9 \text{ Hz}$ ), 202.5; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3067, 3034, 2958, 2929, 2872, 1749, 1589, 1575, 1453, 1261, 1220, 1187, 113, 1100, 1046, 909, 752, 731, 695; MS (ESI+) m/z 356 (100,  $[M + H]^+$ ), 336 (20), 312 (30), 288 (10); HRMS (ESI+) m/z 356.1665 calcd for C<sub>21</sub>H<sub>22</sub>FNO<sub>3</sub> + H<sup>+</sup> 356.1662.

(Z)-3-Amino-3-phenylacrylaldehyde (4). In 10 mL roundbottom flask equipped with a magnetic stirrer bar was added benzyl hydroxy(1-phenyl-3-(trimethylsilyl)prop-2-yn-1-yl)carbamate 10 (178.6 mg, 0.5 mmol, 1 equiv), methanol (3 mL), water (0.25 mL) and potassium carbonate (69 mg, 0.5 mmol, 1 equiv), and the resulting mixture was stirred at room temperature for 3 h. The reaction mixture was then filtered through a silica gel pad and eluted with ethyl acetate. The filtrate was concentrated, and the residue was purified by flash chromatography on silica gel (pentane/ethyl acetate  $8/2 \rightarrow 7/3$ ,  $R_f$ (silica, pentane/ethyl acetate 8/2 = 0.3, UV/Vanillin), to give 51.8 mg (0.35 mmol, 68% yield) of the title compound 4 as a yellow semisolid: <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 5.40 (s, 1H), 5.78 (brs, 1H), 7.39-7.50 (m, 3H), 7.53-7.59 (m, 2H), 10.00 (brs, 1H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 95.5, 126.2, 128.9, 130.9, 136.2, 162.3, 188.1; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3315, 3165, 3061, 2806, 2728, 1636, 1611, 1576, 1531, 1486, 1449, 1412, 1365, 1298, 1141, 1119, 1075, 996,

918, 773, 729, 692, 623, 611; MS (ESI+) *m/z* 260 (10), 148 (100, [M + H]<sup>+</sup>); HRMS (ESI+) *m/z* 148.0746 calcd for C<sub>0</sub>H<sub>0</sub>NO + H<sup>+</sup> 148.0762.

(**2**)-Benzyl (**3**-oxo-1-phenylbut-1-en-1-yl)carbamate (2p). According to general procedure B (on 0.2 mmol scale) and starting from benzyl (1-phenylbut-2-yn-1-yl)(hydroxy)carbamate 1p, 44.1 mg of compound 2p (0.149 mmol, 75% yield) were obtained as a pale yellow solid after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 92/8):  $R_f$  (silica, Pentane/Et<sub>2</sub>O 9/1 = 0.35, UV/PMA/Vanillin); mp 62–64 °C (Et<sub>2</sub>O/pentane); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 2.23 (s, 3H), 5.08 (s, 2H), 5.59 (s, 1H), 7.27–7.48 (m, 10H), 11.54 (brs, 1H, NH); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 30.8, 67.4, 107.4, 127.4, 127.9, 128.2, 128.3, 128.4, 129.8, 135.3, 135.4, 152.7, 154.5, 199.5; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3033, 2949, 1752, 1639, 1605, 1587, 1570, 1494, 1467, 1454, 1357, 1290, 1206, 1170, 1114, 1046, 766, 747, 694; MS (ESI+) m/z 296 (100, [M + H]<sup>+</sup>), 252 (50), 162 (20); HRMS (ESI+) m/z 296.1286 calcd for C<sub>18</sub>H<sub>17</sub>NO<sub>3</sub> + H<sup>+</sup> 296.1287.

(Z)-Benzyl (4-methyl-3-oxo-1-phenylpent-1-en-1-yl)carbamate (2q). According to general procedure B (on 0.1 mmol scale) and starting from benzyl (4-methyl-1-phenylpent-2-yn-1-yl)-(hydroxy)carbamate 1q, 20.8 mg of compound 2q (0.064 mmol, 63% yield) were obtained as a pale yellow oil after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 92/8).  $R_f$  (silica, Pentane/  $Et_2O 9/1 = 0.4$ , UV/PMA/Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$ (ppm) 1.15 (d, J = 7.2 Hz, 6H), 2.65 (sept., J = 7.2 Hz, 1H), 5.08 (s, 2H), 5.61 (s, 1H), 7.27-7.48 (m, 10H), 11.61 (brs, 1H, NH); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 17.7, 40.3, 66.4, 104.9, 126.5, 127.0, 127.2, 127.3, 127.5, 128.7, 135.5, 135.7, 151.8, 154.1, 205.1; IR (ATR, neat) v (cm<sup>-1</sup>) 3065, 3034, 2969, 2872, 1749, 1639, 1604, 1587, 1573, 1495, 1462, 1286, 1194, 1174, 1129, 1069, 1048, 906, 726, 694; MS (ESI+) m/  $z 647 (5, [2M + H]^+), 324 (100, [M + H]^+), 306 (50), 280 (45), 262$ (10); HRMS (ESI+) m/z 324.1599 calcd for  $C_{20}H_{21}NO_3 + H^+$ 324.1600.

(*E*)-Benzyl (4,4-dimethyl-1-phenylpent-2-yn-1-ylidene)carbamate (3r). According to general procedure B (on 0.2 mmol scale) and starting from benzyl (4,4-dimethyl-1-phenylpent-2-yn-1yl)(hydroxy)carbamate 1r, 25.3 mg of compound 3r (0.079 mmol, 41% yield) were obtained as a yellow oil after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 95/5):  $R_f$  (silica, Pentane/Et<sub>2</sub>O 9/1 = 0.45, UV/PMA); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 1.28 (s, 9H), 5.31 (s, 2H), 7.30–7.55 (m, 8H), 8.06 (~d, *J* = 7.2 Hz, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 28.3, 30.1, 68.3, 72.8, 110.7, 128.3, 128.35, 128.4, 128.5, 128.6, 132.4, 135.0, 135.5, 153.6, 163.0; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3065, 3033, 2970, 2929, 2868, 2218, 1722, 1595, 1573, 1450, 1318, 1295, 1256, 1201, 1176, 905, 690.MS (ESI+) *m*/*z* 320 (100, [M + H]<sup>+</sup>), 276 (5), 228 (10); HRMS (ESI+) *m*/ *z* 320.1638 calcd for C<sub>21</sub>H<sub>21</sub>NO<sub>2</sub> + H<sup>+</sup> 320.1651.

(Z)-Benzyl (3-oxo-1,3-diphenylprop-1-en-1-yl)carbamate (2s) and (E)-Benzyl (1,3-diphenylprop-2-yn-1-ylidene)carbamate (3s). According to general procedure B (on 0.2 mmol scale) and starting from benzyl (1,3-diphenylprop-2-yn-1-yl)(hydroxy)carbamate 1s, 36.3 mg of compound 2s as a viscous yellow oil (0.101 mmol, 51% yield) and 12.0 mg of compound 3s as a pale yellow oil (0.035 mmol, 18% yield) were obtained. **2s**,  $R_f$  (silica, Pentane/Et<sub>2</sub>O 9/ 1) = 0.45, UV/PMA/Vanillin), and 3s,  $R_f$  (silica, Pentane/Et<sub>2</sub>O 9/1) = 0.4, UV/PMA), were separated by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 95/5). **2s**: <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 5.11 (s, 2H), 6.29 (s, 1H), 7.28–7.58 (m, 13H), 7.93–7.98 (m, 2H), 12.01 (brs, 1H, NH); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 67.5, 103.8, 127.6, 127.8, 128.1, 128.3, 128.4, 128.5, 128.6, 129.9, 132.6, 135.4, 136.0, 138.6, 152.8, 156.8, 191.3; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3063, 3032, 2923, 2851, 1749, 1722, 1621, 1586, 1567, 1496, 1469, 1319, 1291, 1171, 1126, 1045, 1020, 776, 750, 688; MS (ESI+) m/z 358 (100,  $[M + H]^+$ ), 340 (40), 314 (15), 288 (15); HRMS (ESI+) m/z 358.1447 calcd for  $C_{23}H_{19}NO_3 + H^+ 358.1443.$  3s: <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 5.38 (s, 2H), 7.27-7.34 (m, 2H), 7.34-7.40 (m, 3H), 7.41-7.50 (m, 7H), 7.53-7.58 (m, 1H), 8.16-8.19 (m, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 68.4, 81.7, 100.1, 120.3, 128.3, 128.4, 128.5, 128.7, 130.5, 132.6, 132.7, 134.9, 135.6, 153.3, 163.0; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3083, 3060, 2215, 2192, 1711, 1589, 1566, 1490, 1446, 1330, 1255, 1239, 1195, 1172, 776, 752, 699, 690, 678, 605; MS (ESI+) m/z 679 (20,

 $[2M + H]^+$ ), 340 (100,  $[M + H]^+$ ), 169 (5); HRMS (ESI+) *m/z* 340.1333 calcd for C<sub>23</sub>H<sub>17</sub>NO<sub>2</sub> + H<sup>+</sup> 340.1338.

(Z)-Benzyl (3-oxo-1-phenyl-3-(p-tolyl)prop-1-en-1-yl)carbamate (2t) and (E)-Benzyl (1-phenyl-3-(p-tolyl)prop-2-yn-1-ylidene)carbamate (3t). According to general procedure B (on 0.2 mmol scale) and starting from benzyl (1-phenyl-3-(p-tolyl)prop-2-yn-1yl)(hydroxy)carbamate 1t, 39.2 mg of compound 2t as a pale yellow oil (0.105 mmol, 53% yield) and 11.8 mg of compound 3t as a colorless oil (0.033 mmol, 14% yield) were obtained. **2t**,  $R_f$  (silica, Pentane/Et<sub>2</sub>O 9/ 1) = 0.45, UV/PMA/Vanillin), and 3t,  $R_f$  (silica, Pentane/Et<sub>2</sub>O 9/1) = 0.4, UV/PMA), were separated by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 95/5). 2t: <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ (ppm) 2.41 (s, 3H), 5.11 (s, 2H), 6.28 (s, 1H), 7.26 (d, J = 8.0 Hz, 2H), 7.30-7.38 (m, 5H), 7.40–7.52 (m, 5H), 7.86 (d, J = 8.4 Hz, 2H), 12.03 (brs, 1H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 21.6, 67.5, 103.8, 127.6, 127.9, 128.0, 128.3, 128.5, 129.3, 129.8, 135.5, 136.0, 136.1, 143.4, 152.8, 156.3, 191.0; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3061, 3034, 2957, 1748, 1622, 1605, 1585, 1572, 1495, 1472, 1315, 1294, 1190, 1173, 1044, 805, 732, 694; MS (ESI+) m/z 372 (100,  $[M + H]^+$ ), 354 (45); HRMS (ESI+) m/z372.1599 calcd for C<sub>24</sub>H<sub>21</sub>NO<sub>3</sub> + H<sup>+</sup> 372.1600. 3t: <sup>1</sup>H NMR (400 MHz,  $CDCl_3$ )  $\delta$  (ppm) 2.40 (s, 3H), 5.38 (s, 2H), 7.18(d, J = 8 Hz, 2H), 7.28-7.36 (m, 5H), 7.44-7.50 (m, 4H), 7.52-7.57 (m, 1H), 8.15-8.19 (m, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ (ppm) 21.7, 68.4, 81.5, 100.8, 117.2, 128.3, 128.4, 128.5, 128.6, 128.7, 129.3, 132.6, 135.0, 135.6, 141.2, 153.4, 163.0; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3064, 3032, 2953, 2207, 2189, 1718, 1593, 1569, 1508, 1449, 1324, 1232, 1193, 1171, 1048, 814, 688; MS (ESI+) m/z 354 (100,  $[M + H]^+$ ); HRMS (ESI+) m/z 354.1505 calcd for  $C_{24}H_{19}NO_2 + H^+$  354.1494.

(Z)-1-Amino-1-phenylhept-1-en-3-one (7). In 10 mL roundbottom flask equipped with a magnetic stirrer bar was added (Z)-benzyl (3-oxo-1-phenylhept-1-en-1-yl)carbamate 2a (67.5 mg, 0.2 mmol, 1 equiv), methanol (2 mL), palladium on carbon (wt. 10%, 23 mg, 0.02 mmol, 0.1 equiv), and the resulting mixture was stirred at room temperature under hydrogen atmosphere for 30 min. The reaction mixture was then filtered through a silica gel pad and eluted with ethyl acetate. The filtrate was concentrated, and the residue was purified by flash chromatography on silica gel (pentane/ethyl acetate 8/2,  $R_f$  (silica, pentane/ethyl acetate 8/2 = 0.25, UV/Ninhydrin)) to give 39.1 mg (0.19 mmol, 97% yield) of the title compound 7 as a pale yellow oil: <sup>1</sup>H NMR (400 MHz,  $CDCl_3$ )  $\delta$  (ppm) 0.92 (t, J = 7.6 Hz, 3H), 1.36 (sext, J = 7.6 Hz, 2H), 1.63 (pent, J = 7.6 Hz, 2H), 2.38 (t, J = 7.6 Hz, 2H), 5.25 (brs, 1H), 5.44 (s, 1H), 7.38-7.48 (m, 3H), 7.38-7.48 (m, 2H), 9.95 (brs, 1H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ (ppm) 13.9, 22.6, 28.0, 42.6, 94.7, 126.2, 128.8, 130.4, 137.4, 160.8, 200.7. The spectral data are consistent with those of the literature.

**4-Butyl-2,6-diphenylpyrimidine (6).** According to general procedure C (on 0.1 mmol scale) and starting from (*Z*)-benzyl (3-oxo-1-phenylhept-1-en-1-yl)carbamate **2a** and benzamide, 23.0 mg of compound **6** (0.797 mmol, 80% yield) were obtained as a white solid after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 98/2):  $R_f$  (silica, pentane/Et<sub>2</sub>O 95/5 = 0.8, UV/Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 1.01 (t, *J* = 7.4 Hz, 3H), 1.43–1.54 (m, 2H), 1.82–1.91 (m, 2H), 2.90 (t, *J* = 7.4 Hz, 2H), 7.46 (s, 1H), 7.47–7.57 (m, 6H), 8.21–8.26 (m, 2H), 8.63 (dd, *J* = 1.8 Hz and *J* = 7.8 Hz, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 14.0, 22.5, 31.0, 37.9, 113.4, 127.2, 128.3, 128.4, 128.8, 130.4, 130.6, 137.4, 138.1, 163.7, 164.2, 171.6. The spectral data are consistent with those of the literature.<sup>28</sup>

**4-Butyl-2-phenyl-6-(***p***-tolyl)pyrimidine (9).** According to general procedure C (on 0.2 mmol scale) and starting from (*Z*)-benzyl (3-oxo-1-(*p*-tolyl)hept-1-en-1-yl)carbamate **2b** and benzamide, 53.4 mg of compound **9** (0.177 mmol, 89% yield) were obtained as a pale yellow solid after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 98/2):  $R_f$  (silica, pentane/Et<sub>2</sub>O 95/5 = 0.8, UV/Vanillin); mp 61–63 °C (Et<sub>2</sub>O/Pentane); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 1.01 (t, *J* = 7.6 Hz, 3H), 1.44–1.54 (m, 2H), 1.82–1.91 (m, 2H), 2.45 (s, 3H), 2.88 (t, *J* = 7.6 Hz, 2H), 7.34 (d, *J* = 8 Hz, 2H) 7.43 (s, 1H), 7.47–7.56 (m, 3H), 8.14 (d, *J* = 8 Hz, 2H), 8.63 (dd, *J* = 1.7 Hz and *J* = 7.9 Hz, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 13.9, 21.4, 22.5, 31.0, 37.9, 113.0, 127.1, 128.3, 129.5, 130.3, 134.6, 138.4, 140.8, 163.6, 164.1, 171.4; IR (ATR, neat) ν (cm<sup>-1</sup>) 2956, 2929, 1588, 1571, 1528, 1511,

14818, 1360, 820, 752, 695, 661; MS (ESI+) m/z 303 (100, [M + H]<sup>+</sup>); HRMS (ESI+) m/z 303.1853 calcd for C<sub>21</sub>H<sub>22</sub>N<sub>2</sub> + H<sup>+</sup> 303.1856.

4-Butyl-2-phenyl-6-(m-tolyl)pyrimidine (10). According to general procedure C (on 0.2 mmol scale) and starting from (Z)-benzyl (3-oxo-1-(*m*-tolyl)hept-1-en-1-yl)carbamate **2g** and benzamide, 49.0 mg of compound 10 (0.162 mmol, 81% yield) were obtained as a pale yellow oil after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 98/2): R<sub>f</sub> (silica, pentane/Et<sub>2</sub>O 95/5 = 0.9, UV/ Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 1.01 (t, J = 7.4 Hz, 3H), 1.43–1.58 (m, 2H), 1.82–1.91 (m, 2H), 2.49 (s, 3H), 2.89 (t, J = 7.6 Hz, 2H), 7.33 (d, J = 7.6 Hz, 1H) 7.42 (t, J = 7.6 Hz, 1H), 7.45 (s, 1H), 7.48–7.56 (m, 3H), 8.00 (d, J = 7.6 Hz, 1H), 8.05 (brs, 1H), 8.62 (dd, J = 1.8 Hz and J = 7.9 Hz, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$ (ppm) 14.0, 21.5, 22.5, 31.0, 38.0, 113.5, 124.3, 127.8, 128.4, 128.7, 130.3, 131.3, 137.4, 138.3, 138.5, 163.9, 164.2, 171.5; IR (ATR, neat) v (cm<sup>-1</sup>) 2955, 2926, 2870, 2859, 1587, 1569, 1532, 1367, 1173, 1027, 755, 692, 666, 639; MS (ESI+) m/z 303 (100,  $[M + H]^+$ ); HRMS (ESI +) m/z 303.1861 calcd for C<sub>21</sub>H<sub>22</sub>N<sub>2</sub> + H<sup>+</sup> 303.1861.

4-Butyl-6-(4-fluorophenyl)-2-phenylpyrimidine (11). According to general procedure C (on 0.2 mmol scale) and starting from (Z)benzyl (1-(4-fluorophenyl)-3-oxohept-1-en-1-yl)carbamate 2c and benzamide, 41.0 mg of compound 11 (0.134 mmol, 67% yield) were obtained as a pale yellow solid after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 98/2):  $R_f$  (silica, pentane/Et<sub>2</sub>O 95/5 = 0.75, UV/Vanillin); mp 78-79 °C (Et<sub>2</sub>O/pentane); <sup>1</sup>H NMR (400 MHz,  $CDCl_3$ )  $\delta$  (ppm) 1.00 (t, J = 7.2 Hz, 3H), 1.42–1.53 (m, 2H), 1.81–1.90 (m, 2H), 2.88 (t, J = 7.6 Hz, 2H), 7.21 (~t, J = 8.8 Hz, 2H), 7.40 (s, 1H),7.47-7.56 (m, 3H), 8.17-8.29 (m, 2H), 8.60 (dd, J = 2.4 Hz and J = 8.0Hz, 2H). <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) –110.2; <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 13.9, 22.5, 31.0, 37.9, 113.0, 115.8 (d,  $J_{ortho(C-F)} = 21.7 \text{ Hz}$ ), 128.3, 128.4, 129.2 (d,  $J_{meta(C-F)} = 8.7 \text{ Hz}$ ), 130.5, 133.5 (d,  $J_{para(C-F)} = 3.2$  Hz), 138.9, 157.1, 162.5, 164.2, 164.4 (d,  $J_{inso(C-F)} = 250.9 \text{ Hz}$ , 171.2; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 2957, 2929, 2870, 1603, 1586, 1573, 1527, 1509, 1416, 1360, 1231, 1219, 1163, 831, 751, 690, 663; MS (ESI+) m/z 307 (100,  $[M + H]^+$ ); HRMS (ESI+) m/z307.1608 calcd for  $C_{20}H_{19}FN_2 + H^+ 307.1611$ .

4-Butyl-6-(2-fluorophenyl)-2-phenylpyrimidine (12). According to general procedure C (on 0.2 mmol scale) and starting from (Z)benzyl (1-(2-fluorophenyl)-3-oxohept-1-en-1-yl)carbamate 2n and benzamide, 39.1 mg of compound 12 (0.128 mmol, 64% yield) were obtained as a viscous colorless oil after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 98/2): R<sub>f</sub> (silica, pentane/Et<sub>2</sub>O 95/5 = 0.8, UV/Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 1.00 (t, J = 7.2 Hz, 3H, H<sub>1</sub>), 1.43–1.53 (m, 2H,  $H_2$ ), 1.81–1.90 (m, 2H,  $H_3$ ), 2.90 (t, J = 7.6 Hz, 2H,  $H_4$ ), 7.20 (ddd, J = 1.0 Hz, J = 8.3 Hz, J = 11.7 Hz, 1H, H<sub>Ar</sub>), 7.34 (dt, J = 1.1 Hz and J = 7.6Hz, 1H,  $H_{Ar}$ ), 7.43–7.55 (m, 4H,  $H_{Ar}$ ), 7.59 (d, J = 2.1 Hz, 1H,  $H_{Ar}$ ), 8.36 (dt, J = 1.9 Hz and J = 7.8 Hz, 1H, H<sub>Ar</sub>), 8.57–8.61 (m, 2H, H<sub>16</sub>). <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>) δ (ppm) –114.8; <sup>13</sup>C NMR (100 MHz,  $CDCl_3$ )  $\delta$  (ppm) 13.9, 22.5, 31.0, 37.9, 116.3 (d,  $J_{ortho(C-F)} = 22.9 \text{ Hz}$ ), 117.7 (d,  $J_{meta(C-F)} = 8.7$  Hz), 124.6 (d,  ${}^{4}J_{(C-F)} = 3.5$  Hz), 125.5 (d,  ${}^{4}J_{(C-F)}$  = 11.5 Hz), 128.3, 128.4, 130.4, 131.1 (d,  $J_{para(C-F)}$  = 2.5 Hz), 131.8 (d,  $J_{meta(C-F)} = 8.8$  Hz), 138.1, 159.7 (d,  $J_{(C-F)} = 2.5$  Hz), 161.4 (d,  $J_{i\nu so(C-F)} = 251.9 \text{ Hz}$ , 164.1, 171.6; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3074, 3036, 2959, 2926, 2873, 2857, 1614, 1568, 1527, 1489, 1373, 1358, 1260, 1209, 872, 853, 773, 753, 734, 726, 694, 628; MS (ESI+) m/z 307 (100,  $[M + H]^+$ ; HRMS (ESI+) m/z 307.1606 calcd for  $C_{20}H_{19}FN_2 + H^+$ 307.1611.

**4-(3-Bromophenyl)-6-butyl-2-phenylpyrimidine (13).** According to general procedure C (on 0.2 mmol scale) and starting from (*Z*)-benzyl (1-(3-bromophenyl)-3-oxohept-1-en-1-yl)carbamate **2i** and benzamide, 39.0 mg of compound **13** (0.106 mmol, 54% yield) were obtained as a pale yellow oil after purification by flash chromatography on silica gel (petroleum ether/Et<sub>2</sub>O 97/3):  $R_f$  (silica, pentane/Et<sub>2</sub>O 95/ S = 0.8, UV/Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 1.01 (t, *J* = 7.6 Hz, 3H), 1.42–1.52 (m, 2H), 1.79–1.89 (m, 2H), 2.89 (t, *J* = 7.6 Hz, 2H), 7.35–7.66 (m, 6H), 8.09–8.15 (m, 1H), 8.31–8.40 (m, 1H), 8.50–8.75 (m, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 13.9, 22.5, 30.9, 38.0, 113.4, 123.1, 125.7, 128.3, 128.4, 130.2, 130.3, 130.5, 133.4, 138.0, 139.5, 162.1, 164.3, 172.0; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3066, 2957,

2928, 2859, 1587, 1563, 1531, 1367, 1115, 1069, 789, 756, 689, 660, 634; MS (ESI+) m/z 369 (98,  $[M + H]^{+,81}Br$ ), 367 (100,  $[M + H]^{+,79}Br$ ). 303 (15); HRMS (ESI+) m/z 367.0815 calcd for  $C_{20}H_{19}BrN_2$  + H<sup>+</sup> 367.0810.

4-Butyl-6-(3-methoxyphenyl)-2-phenylpyrimidine (14). According to general procedure C (on 0.2 mmol scale) and starting from (Z)-benzyl (1-(3-methoxyphenyl)-3-oxohept-1-en-1-yl)carbamate 2h and benzamide, 38.9 mg of compound 14 (0.122 mmol, 62% yield) were obtained as a pale yellow oil after purification by flash chromatography on silica gel (petroleum ether/Et<sub>2</sub>O 97/3): R<sub>f</sub> (silica, pentane/Et<sub>2</sub>O 95/5 = 0.6, UV/Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 1.01 (t, J = 7.6 Hz, 3H), 1.43–1.54 (m, 2H), 1.82–1.91 (m, 2H), 2.89 (t, J = 7.6 Hz, 2H), 3.94 (s, 3H), 7.07 (dd, J = 2.5 Hz and J = 8.2 Hz, 1H), 7.42-7.47 (m, 2H), 7.48-7.56 (m, 3H), 7.77 (d, J = 7.7 Hz, 1H), 7.84 (brs, 1H), 8.60-8.65 (m, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 13.9, 22.5, 30.9, 37.9, 55.4, 112.6, 113.5, 116.1, 119.5, 128.3, 128.4, 129.8, 130.4, 138.2, 138.9, 160.1, 163.4, 164.1, 171.6; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3066, 2955, 2930, 2870, 1587, 1568, 1531, 1463, 1368, 1253, 1045, 858, 756, 691, 673, 636; MS (ESI+) m/z 319 (100, [M +  $H^{+}$ ; HRMS (ESI+) m/z 319.1805 calcd for  $C_{21}H_{22}N_2O$  +  $H^{+}$ 319.1810.

**4-Butyl-6-(naphthalen-2-yl)-2-phenylpyrimidine (15).** According to general procedure C (on 0.2 mmol scale) and starting from (*Z*)-benzyl (1-(naphthalen-2-yl)-3-oxohept-1-en-1-yl)carbamate **2k** and benzamide, 38.2 mg of compound **15** (0.113 mmol, 57% yield) were obtained as a yellow oil after purification by flash chromatography on silica gel (petroleum ether/Et<sub>2</sub>O 98/2):  $R_f$  (silica, pentane/Et<sub>2</sub>O 95/5 = 0.8, UV/Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 1.04 (t, *J* = 7.6 Hz, 3H), 1.46–1.57 (m, 2H), 1.85–1.95 (m, 2H), 2.93 (t, *J* = 8.0 Hz, 2H), 7.50–7.62 (m, 6H), 7.89–8.05 (m, 3H), 8.35 (d, *J* = 8.4 Hz, 1H), 8.66–8.74 (m, 3H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 14.0, 22.5, 31.0, 38.0, 113.6, 124.2, 126.4, 127.1,127.2, 127.7, 128.3, 128.4, 128.5, 128.9, 130.4, 133.3, 134.5, 134.7, 138.3, 163.5, 164.3, 171.6; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3060, 2954, 2927, 2858, 1587, 1568, 1531, 1373; 1173, 854, 817, 758, 694, 656; MS (ESI+) *m*/*z* 339 (100, [M + H]<sup>+</sup>); HRMS (ESI+) *m*/*z* 339.1861 calcd for C<sub>24</sub>H<sub>22</sub>N<sub>2</sub> + H<sup>+</sup> 339.1861.

4-Butyl-2-phenyl-6-(thiophen-2-yl)pyrimidine (16). According to general procedure C (on 0.2 mmol scale) and starting from (Z)benzyl (3-oxo-1-(thiophen-2-yl)hept-1-en-1-yl)carbamate 2j and benzamide, 30.5 mg of compound 16 (0.104 mmol, 52% yield) were obtained as a yellow oil after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 98/2):  $R_f$  (silica, pentane/Et<sub>2</sub>O 95/5 = 0.7, UV/Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 1.00 (t, J = 7.2 Hz, 3H), 1.42–1.52 (m, 2H), 1.79–1.88 (m, 2H), 2.85 (t, J = 8 Hz, 2H), 7.17 (dd, J = 3.9 Hz and J = 4.8 Hz, 1H), 7.30 (s, 1H), 7.44-7.56 (m, 4H), 7.82 (d, J = 3.2 Hz, 1H), 8.57 (dd, J = 1.9 Hz and J = 7.6 Hz, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 13.9, 22.5, 30.9, 37.8, 111.5, 126.8, 128.1, 128.3, 128.4, 129.5, 130.4, 137.8, 143.3, 158.7, 164.1, 171.4; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 2955, 2927, 2858, 1720, 1588, 1570, 1528, 1432, 1373, 1269, 1235, 1172, 1026, 856, 756, 694; MS (ESI+) m/ z 295 (100,  $[M + H]^+$ ); HRMS (ESI+) m/z 295.1261 calcd for  $C_{18}H_{18}N_2S + H^+ 295.1269.$ 

**4-Methyl-2,6-diphenylpyrimidine (17).** According to general procedure C (on 0.2 mmol scale) and starting from (*Z*)-benzyl (3-oxo-1-phenylbut-1-en-1-yl)carbamate **2p** and benzamide, 32.9 mg of compound **17** (0.134 mmol, 77% yield) were obtained as a pale yellow solid after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 96/4): *R*<sub>f</sub> (silica, pentane/Et<sub>2</sub>O 95/5 = 0.4, UV/Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 2.65 (s, 3H), 7.47 (s, 1H), 7.49–7.57 (m, 6H), 8.20–8.25 (m, 2H), 8.58–8.64 (m, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 24.6, 114.0, 127.1, 128.3, 128.4, 128.8, 130.4, 130.6, 137.2, 138.1, 163.6, 164.2, 167.7. The spectral data are consistent with those of the literature.<sup>29</sup>

**4-Isopropyl-2,6-diphenylpyrimidine (18).** According to general procedure C (on 0.2 mmol scale) and starting from (*Z*)-benzyl (4-methyl-3-oxo-1-phenylpent-1-en-1-yl)carbamate **2q** and benzamide, 39.2 mg of compound **18** (0.142 mmol, 71% yield) were obtained as a pale yellow oil after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 98/2):  $R_f$  (silica, pentane/Et<sub>2</sub>O 95/5 = 0.8, UV/ Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 1.44 (t, *J* = 6.8 Hz, 6H,

H<sub>1</sub> and H<sub>1</sub>'), 3.15 (sept., *J* = 6.8 Hz, 1H, H<sub>2</sub>), 7.48 (s, 1H, H<sub>4</sub>), 7.49–7.58 (m, 6H, H<sub>Ar</sub>), 8.24 (dd, *J* = 2.0 Hz and *J* = 8.0 Hz, 2H, H<sub>7</sub>), 8.66 (dd, *J* = 2.0 Hz and *J* = 8.0 Hz, 2H, H<sub>1</sub>); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ (ppm) 21.9, 36.3, 111.5, 127.2, 128.4, 128.8, 130.3, 130.5, 137.6, 138.3, 163.9, 164.0, 176.3; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 2961, 2924, 2866, 1590, 1568, 1531, 1496, 1373, 756, 692, 633; MS (ESI+) *m*/*z* 275 (100, [M + H]<sup>+</sup>); HRMS (ESI+) *m*/*z* 275.1551 calcd for C<sub>19</sub>H<sub>18</sub>N<sub>2</sub> + H<sup>+</sup> 275.1548.

**2,4,6-Triphenylpyrimidine (19).** According to general procedure C (on 0.1 mmol scale) and starting from (*Z*)-benzyl (3-oxo-1,3-diphenylprop-1-en-1-yl)carbamate **2s** and benzamide, 16.6 mg of compound **19** (0.054 mmol, 54% yield) were obtained as a white solid after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 98/2):  $R_f$  (silica, pentane/Et<sub>2</sub>O 95/5 = 0.4, UV/Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 7.50–7.62 (m, 9H), 8.06 (s, 1H), 8.28–8.33 (dd, 1H, J = 1.6 Hz and J = 8.0 Hz, 4H), 8.28–8.33 (dd, 1H, J = 2.0 Hz and J = 8.0 Hz, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 110.3, 127.3,128.4, 128.9, 130.6, 130.7, 137.5, 138.1, 164.5, 164.7. The spectral data are consistent with those of the literature.<sup>30</sup>

**4-Butyl-6-phenyl-2-(thiophen-2-yl)pyrimidine (20).** According to general procedure C (on 0.2 mmol scale) and starting from (*Z*)-benzyl (3-oxo-1-phenylhept-1-en-1-yl)carbamate **2a** and 2-thiophene-carboxamide, 36.2 mg of compound **20** (0.122 mmol, 60% yield) were obtained as a pale yellow oil after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 98/2):  $R_f$  (silica, pentane/Et<sub>2</sub>O 95/5 = 0.75, UV/Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 1.00 (t, *J* = 7.6 Hz, 3H), 1.42–1.52 (m, 2H), 1.78–1.88 (m, 2H), 2.84 (t, *J* = 7.6 Hz, 2H), 7.17 (dd, *J* = 3.7 Hz and *J* = 5.0 Hz, 1H), 7.37 (s, 1H), 7.48 (dd, *J* = 1.2 Hz and *J* = 3.6 Hz, 1H), 8.17–8.21 (m, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 13.9, 22.4, 30.8, 37.7, 112.8, 127.1, 128.0, 128.6, 128.8, 129.3, 130.6, 137.0, 144.2, 161.1, 163.6, 171.7. The spectral data are consistent with those of the literature.<sup>31</sup>

**4-Butyl-2-(4-nitrophenyl)-6-phenylpyrimidine (21).** According to general procedure C (on 0.2 mmol scale) and starting from (*Z*)-benzyl (3-oxo-1-phenylhept-1-en-1-yl)carbamate **2a** and 4-nitrobenzamide, 54.6 mg of compound **21** (0.164 mmol, 82% yield) were obtained as a yellow solid after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 97/3):  $R_f$  (silica, pentane/Et<sub>2</sub>O 95/5 = 0.5, UV/Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 1.01 (t, *J* = 7.2 Hz, 3H), 1.43–1.54 (m, 2H), 1.80–1.90 (m, 2H), 2.90 (t, *J* = 7.6 Hz, 2H), 7.52–7.58 (m, 4H), 8.18 – 8.22 (m, 2H), 8.33 (~d, *J* = 8.8 Hz, 2H),8.77 (~d, *J* = 9.2 Hz, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 13.9, 22.5, 30.9, 37.9, 114.4, 123.5, 127.2, 128.9, 129.2, 131.0, 136.8, 144.0, 149.1, 162.0, 164.0, 172.1; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 2953, 2926, 2869, 1577, 1520, 1463, 1420, 1353, 1336, 1223, 867, 857, 829, 773, 743, 690, 646; MS (ESI+) *m*/z 334.1536 calcd for C<sub>20</sub>H<sub>19</sub>N<sub>3</sub>O<sub>2</sub> + H<sup>+</sup> 334.1556.

2-(4-Bromophenyl)-4-butyl-6-phenylpyrimidine (22). According to general procedure C (on 0.2 mmol scale) and starting from (Z)benzyl (3-oxo-1-phenylhept-1-en-1-yl)carbamate 2a and 4-bromobenzamide, 56.7 mg of compound 22 (0.154 mmol, 78% yield) were obtained as a pale yellow solid after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 98/2):  $R_f$  (silica, pentane/Et<sub>2</sub>O 95/5 = 0.7, UV/Vanillin); mp 88-90 °C (Et<sub>2</sub>O/pentane); <sup>1</sup>H NMR (400 MHz,  $CDCl_3$ )  $\delta$  (ppm) 1.00 (t, J = 7.6 Hz), 1.42–1.53 (m, 2H), 1.79–1.88 (m, 2H), 2.88 (t, J = 7.6 Hz, 2H), 7.46 (s, 1H), 7.50-7.56 (m, 3H), 7.63 (~d, J = 8.8 Hz, 2H), 8.18–8.22 (m, 2H), 8.49 (~d, J = 8.8 Hz, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ (ppm) 13.9, 22.5, 31.0, 37.9, 113.6, 125.1, 127.2, 128.8, 130.0, 130.7, 131.5, 137.1, 137.2, 163.3, 163.8, 171.7; IR (ATR, neat)  $\nu$  (cm  $^{-1}$ ) 2956, 2924, 2857, 1588, 1564, 1530, 1363, 1169, 1069, 1010, 841, 767, 690, 639; MS (ESI+) m/z 369 (97, [M + H]<sup>+,81</sup>Br), 367 (100, [M + H]<sup>+,79</sup>Br). 288 (15), 275 (10); HRMS (ESI +) m/z 367.0796 calcd for  $C_{20}H_{19}BrN_2 + H^+$  367.0810.

**4-Butyl-6-phenyl-2-vinylpyrimidine (23).** According to general procedure D (on 0.1 mmol scale) and starting from (*Z*)-benzyl (3-oxo-1-phenylhept-1-en-1-yl)carbamate **2a** and acrylamide, 17.1 mg of compound **23** (0.072 mmol, 72% yield) were obtained as a yellow oil after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 95/5):  $R_f$  (silica, pentane/Et<sub>2</sub>O 9/1 = 0.6, UV/Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 0.97 (t, *J* = 7.4 Hz, 3H), 1.38–1.49 (m, 2H),

1.73–1.83 (m, 2H), 2.81 (t, *J* = 7.6 Hz, 2H), 5.74 (dd, *J* = 1.9 Hz and *J* = 10.5 Hz, 1H), 6.75 (dd, *J* = 1.9 Hz and *J* = 17.3 Hz, 1H), 6.96 (dd, *J* = 10.5 Hz and *J* = 17.3 Hz, 1H), 7.40 (s, 1H), 7.47–7.53 (m, 3H), 8.09–8.14 (m, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 13.9, 22.5, 31.1, 37.8, 113.6, 123.5, 127.2, 128.8, 130.6, 137.0, 137.2, 163.8, 164.0, 171.3; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 2956, 2926, 2857, 1573, 1531, 1420, 1371, 991, 940, 772, 690; MS (ESI+) *m*/*z* 239 (100, [M + H]<sup>+</sup>); HRMS (ESI+) *m*/*z* 239.1537 calcd for C<sub>16</sub>H<sub>18</sub>N<sub>2</sub> + H<sup>+</sup> 239.1548.

4-Butyl-6-(4-fluorophenyl)-2-vinylpyrimidine (24). According to general procedure D (on 0.1 mmol scale) and starting from (Z)benzyl (1-(4-fluorophenyl)-3-oxohept-1-en-1-yl)carbamate 2c and acrylamide, 17.7 mg of compound 24 (0.069 mmol, 69% yield) were obtained as a yellow oil after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 96/4):  $R_f$  (silica, pentane/Et<sub>2</sub>O 9/1 = 0.7, UV/ Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 0.97 (t, J = 7.3 Hz, 3H,  $H_1$ ), 1.38–1.49 (m, 2H,  $H_2$ ), 1.72–1.82 (m, 2H,  $H_3$ ), 2.80 (t, J = 7.6 Hz, 2H, H<sub>4</sub>), 5.74 (dd, J = 1.9 Hz and J = 10.4 Hz, 1H, H<sub>14</sub>), 6.73 (dd, J = 2.0Hz and J = 17.3 Hz, 1H, H<sub>14</sub>), 6.94 (dd, J = 10.4 Hz and J = 17.3 Hz, 1H, H<sub>13</sub>), 7.14–7.21 (m, 2H, H<sub>10</sub>), 7.34 (s, 1H, H<sub>6</sub>), 8.09–8.16 (m, 2H, H<sub>9</sub>). <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>) δ (ppm) -110.1; <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 13.9, 22.5, 31.1, 37.8, 113.2, 115.8 (d, *J*<sub>ortho(C-F)</sub> = 21.8 Hz), 123.6, 129.2 (d,  $J_{meta(C-F)} = 8.7$  Hz, 2C), 133.3 (d,  $J_{para(C-F)} = 2.9$ Hz), 136.9, 162.6, 164.0, 164.5 (d,  $J_{ipso(C-F)} = 250.9$  Hz), 171.5; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 2957, 2927, 2858, 1731, 1601, 1575, 1532, 1509, 1232, 1156, 837, 819; MS (ESI+) m/z 420 (15), 257 (100,  $[M + H]^+$ ); HRMS (ESI+) m/z 257.1454 calcd for C<sub>16</sub>H<sub>17</sub>FN<sub>2</sub> + H<sup>+</sup> 257.1454.

**4-Methyl-6-phenyl-2-vinylpyrimidine (25).** According to general procedure D (on 0.1 mmol scale) and starting from (*Z*)-benzyl (3-oxo-1-phenylbut-1-en-1-yl)carbamate **2p** and acrylamide, 13.3 mg of compound **25** (0.0678 mmol, 68% yield) were obtained as a yellow oil after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 88/12):  $R_f$  (silica, pentane/Et<sub>2</sub>O 9/1 = 0.25, UV/Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 2.59 (s, 3H), 5.75 (dd, *J* = 1.9 Hz and *J* = 10.5 Hz, 1H), 6.74 (dd, *J* = 1.9 Hz and *J* = 17.3 Hz, 1H), 6.95 (dd, *J* = 10.5 Hz and *J* = 17.3 Hz, 1H), 7.42 (s, 1H), 7.48–7.53 (m, 3H), 8.09–8.14 (m, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 24.2, 114.3, 123.8, 127.2, 128.9, 130.8, 136.7, 137.0, 163.9, 167.2; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 3034, 2924, 2853, 1574, 1535, 1441, 1354, 773, 690; MS (ESI+) *m*/*z* 197 (100, [M + H]<sup>+</sup>), 184 (5); HRMS (ESI+) *m*/*z* 197.1074 calcd for C<sub>13</sub>H<sub>12</sub>N<sub>2</sub> + H<sup>+</sup> 197.1079.

**4-Butyl-2-methyl-6-phenylpyrimidine (26).** According to general procedure C (on 0.2 mmol scale) and starting from (*Z*)-benzyl (3-oxo-1-phenylhept-1-en-1-yl)carbamate **2a** and acetamide, 31.6 mg of compound **26** (0.142 mmol, 71% yield) were obtained as a pale yellow oil after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 85/15): *R<sub>f</sub>* (silica, pentane/Et<sub>2</sub>O 85/15 = 0.25, UV/Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ (ppm) 0.96 (t, *J* = 7.6 Hz, 3H), 1.37–1.48 (m, 2H), 1.70–1.79 (m, 2H), 2.77 (t, *J* = 7.8 Hz, 2H), 2.78 (s, 3H), 7.35 (s, 1H), 7.46–7.52 (m, 3H), 8.02–8.08 (m, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ (ppm) 13.9, 22.5, 26.2, 31.4, 37.9, 112.8, 127.2, 128.8, 130.5, 137.3, 164.1, 167.9, 171.3; IR (ATR, neat) ν (cm<sup>-1</sup>) 2956, 2926, 2858, 1576, 1537, 1394, 1371, 754, 690; MS (ESI+) *m/z* 227 (100, [M + H]<sup>+</sup>); HRMS (ESI+) *m/z* 227.1527 calcd for C<sub>15</sub>H<sub>18</sub>N<sub>2</sub> + H<sup>+</sup> 227.1548.

**4-Butyl-6-phenylpyrimidine (27).** According to general procedure C (on 0.2 mmol scale) in the absence of 4 Å MS and starting from (*Z*)-benzyl (4-methyl-3-oxo-1-phenylpent-1-en-1-yl)carbamate **2i** and formamide, 20.7 mg of compound **27** (0.098 mmol, 49% yield) were obtained as a yellow oil after purification by flash chromatography on silica gel (pentane/Et<sub>2</sub>O 95/5 → 9/1):  $R_f$  (silica, pentane/Et<sub>2</sub>O 9/1 = 0.3, UV/Vanillin); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 0.96 (t, *J* = 7.2 Hz, 3H), 1.37–1.48 (m, 2H), 1.73–1.82 (m, 2H), 2.82 (t, *J* = 7.6 Hz, 2H), 7.48–7.53 (m, 3H), 7.50 (s, 1H), 8.04–8.10 (m, 2H), 9.10 (s, 1H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm) 13.9, 22.4, 31.1, 37.7, 115.9, 127.1, 128.9, 130.8, 136.8, 158.7, 163.9, 171.3; IR (ATR, neat)  $\nu$  (cm<sup>-1</sup>) 2956, 2927, 2859, 1588, 1578, 1526, 1465, 1442, 1374, 752, 690, 637; MS (ESI+) *m*/*z* 213.1377 calcd for C<sub>14</sub>H<sub>16</sub>N<sub>2</sub> + H<sup>+</sup> 213.1392.

## ASSOCIATED CONTENT

#### **S** Supporting Information

Table of atom coordinates and absolute energies, and NMR spectra (<sup>1</sup>H, <sup>13</sup>C, <sup>19</sup>F) for all products (1a–t, 1v,w, 2a–l, 2n, 2p,q, 2s,t, 4, 6, 7, 9–27). This material is available free of charge via the Internet at http://pubs.acs.org.

#### AUTHOR INFORMATION

#### **Corresponding Author**

\*E-mail: emmanuel.vrancken@enscm.fr; jean-marc.campagne@enscm.fr; helene.gerard@upmc.fr.

#### Notes

The authors declare no competing financial interest.

#### ACKNOWLEDGMENTS

Authors deeply acknowledge Dr. S. Rousseaux and Dr. B. Liegault (ENSCM) for their help in the preparation of this manuscript. We also thank MESR (E. G. Ph.D. grant) and the ENSCM.

## REFERENCES

(1) (a) Elassar, A. A.; El-Khair, A. A. *Tetrahedron* 2003, 59, 8463.
 (b) Lee, D. H.; Park, S.; Cho, K.; Kim, Y.; Athar, T.; Lee, I. *Tetrahedron Lett.* 2007, 48, 8281.
 (c) Xu, X.; Du, P.; Cheng, D.; Wang, H.; Li, X. *Chem. Commun.* 2012, 48, 1811.

(2) Salama, N. N.; Eddington, N. D.; Payne, T. L.; Wilson, K. R.; Scott, K. R. *Curr. Med. Chem.* **2004**, *11*, 2093.

(3) For recent applications and refs cited: (a) Neumann, J. J.; Suri, M.; Glorius, F. Ang. Chem. Int. Ed. 2010, 49, 7790. (b) Zhao, M.; Wang, F.; Li, X. Org. Lett. 2012, 14, 1412. (c) Xiang, D.; Xin, X.; Liu, X.; Zhang, R.; Yang, J.; Dong, D. Org. Lett. 2012, 14, 644. (d) Bunnelle, W. H.; Singman, P. R.; Narayanan, B. A.; Bradshaw, C. W.; Liou, J. S. Synthesis 1997, 439. (e) Yu, R. T.; Rovis, T. J. Am. Chem. Soc. 2006, 128, 12370. (4) (a) Winkler, J. D.; Axten, J. M. J. Am. Chem. Soc. 1998, 120, 6425. (b) Li, G.; Hsung, R. P.; Slafer, B. W.; Sagamanova, I. K. Org. Lett. 2012, 10, 4991. (c) Tsukanov, S. V.; Comins, D. L. Ang. Chem. Int. Ed. 2011, 50, 8626. (d) Caplan, J. F.; Zheng, R.; Blanchard, J. S.; Vederas, J. C. Org. Lett. 2000, 2, 3857. (e) Moreau, J.; Hubert, C.; Batany, J.; Toupet, L.; Roisnel, T.; Hurvois, J.-P.; Renaud, J.-L. J. Org. Chem. 2009, 74, 8963. (f) Edwankar, R. V.; Edwankar, C. R.; Namjoshi, O. A.; Deschamps, J. R.; Cook, J. M. J. Nat. Prod. 2012, 75, 181.

(5) For a recent stereoselective synthesis of Z-enaminones, see ref 1c.
(6) Gayon, E.; Quinonero, O.; Lemouzy, S.; Vrancken, E.; Campagne, J.-M. Org. Lett. 2011, 13, 6418.

(7) (a) Hill, M. D.; Movassaghi, M. Chem.—Eur. J. 2008, 14, 6836.
(b) Michael, J. P. Nat. Prod. Rep. 2005, 22, 627. (c) Erian, A. W. Chem. Rev. 1993, 93, 1991. (d) von Angerer, S. Science of Synthesis; Yamamoto, Y., Ed.; Thieme: Stuttgart, 2003; Vol. 16, pp 379–572. (e) Karpov, A. S.; Müller, T. J. J. Synthesis 2003, 2815.

(8) The use of enaminones in the synthesis of pyrimidines has been described, but the amino group is used as a leaving group: Schenone, P.; Sansebastiano, L.; Mosti, L. *J. Heterocycl. Chem.* **1990**, *27*, 295.

(9) (a) Debleds, O.; Dal Zotto, C.; Vrancken, E.; Campagne, J.-M. *Adv. Synth. Catal.* **2009**, *351*, 1991. (b) Gayon, E.; Debleds, O.; Vrancken, E.; Campagne, J.-M. *J. Org. Chem.* **2010**, *75*, 6050. (c) Debleds, O.; Gayon, E.; Ostaszuk, E.; Vrancken, E.; Campagne, J.-M. *Chem.—Eur. J.* **2010**, 12207. (d) Debleds, O.; Gayon, E.; Vrancken, E.; Campagne, J.-M. *Belstein J. Org. Chem.* **2011**, *7*, 866.

(10) Zhan, Z.-P.; Yu, J.-L.; Liu, H.-J.; Cui, Y.-Y.; Yang, R.-F.; Yang, W.-Z.; Li, J.-P. J. Org. Chem. **2006**, 71, 8298.

(11) See for example: (a) Zanatta, N.; Borchhardt, D. M.; Alves, S. H.; Coehlo, H. S.; Squizani, A. M. C.; Marchi, T. M.; Bonacorso, H. G.; Martins, M. A. P. *Bioorg. Med. Chem.* **2006**, *14*, 3174–3184. (b) Liu, P.; Shan, G.; Chen, S.; Rao, Y. *Tetrahedron Lett.* **2012**, *53*, 936–939.

(12) See for example: (a) Terada, M.; Sorimachi, K. J. Am. Chem. Soc. 2007, 129, 292–293. (b) Terada, M.; Soga, K.; Momiyama, N. Angew. Chem., Int. Ed. 2008, 47, 4122–4125.

(13) The best results are obtained using an  $\ll$  aged  $\gg$  CH<sub>3</sub>CN for which a Karl Fischer titration shows the presence of 0.14% of water, which corresponds to 0.61 equiv compared to the starting propargylic hydroxylamine.

(14) Yoshida, J.-I.; Itoh, M.; Matsunaga, S.-I.; Isoe, S. J. Org. Chem. 1992, 57, 4877 and references therein.

(15) For examples of related eliminations, see: (a) Padwa, A.; Koehler, K. F. *Chem. Commun.* **1986**, 789. (b) Wada, N.; Kaneko, K.; Ukaji, Y.; Inomata, K. *Chem. Lett.* **2011**, 40, 440.

(16) Imine **3a** was isolated as a minor product in a gram scale experiment (70 mg, 4% yield, see Scheme 8, eq 1).

(17) (a) Gilli, P.; Bertolasi, V.; Ferretti, V.; Gilli, G. J. Am. Chem. Soc. 2000, 122, 10405. (b) Sanz, P.; Mó, O.; Yáñez, M.; Elguero, J. J. Phys. Chem. A 2007, 111, 3585.

(18) The best result for the one-pot version of this reaction was obtained when using the following conditions. Starting from 1a: (1) NaOH (10 mol%), CH<sub>3</sub>CN, 50 °C, 1 h followed by the addition in the reaction mixture of (2) *t*-BuOK (2 equiv), PhCONH<sub>2</sub> (1.5 equiv), 4 Å MS, 130 °C, 1 h (sealed tube). 6 was obtained in 23% yield, as compared to the 69% overall yield observed for the stepwise process.

(19) Yamamoto, S.-I.; Okamoto, K.; Murakoso, M.; Kuninobu, Y.; Takai, K. Org. Lett. **2012**, *14*, 3182–3185.

(20) Zhang, Y.; Sheets, M. R.; Raja, E. K.; Boblak, K. N.; Klumpp, D. A. J. Am. Chem. Soc. **2011**, 133, 8467 and refs cited.

(21) Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Mennucci, B.; Petersson, G. A.; Nakatsuji, H.; Caricato, M.; Li, X.; Hratchian, H. P.; Izmaylov, A. F.; Bloino, J.; Zheng, G.; Sonnenberg, J. L.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Montgomery, J. A., Jr.; Peralta, J. E.; Ogliaro, F.; Bearpark, M.; Heyd, J. J.; Brothers, E.; Kudin, K. N.; Staroverov, V. N.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Rega, N.; Millam, J. M.; Klene, M.; Knox, J. E.; Cross, J. B.; Bakken, V.; Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R. E.; Yazyev, O.; Austin, A. J.; Cammi, R.; Pomelli, C.; Ochterski, J. W.; Martin, R. L.; Morokuma, K.; Zakrzewski, V. G.; Voth, G. A.; Salvador, P.; Dannenberg, J. J.; Dapprich, S.; Daniels, A. D.; Farkas, Ö.; Foresman, J. B.; Ortiz, J. V.; Cioslowski, J.; Fox, D. J. Gaussian 09, Revision A.2; Gaussian, Inc.: Wallingford, CT, 2009

(22) Korth, M.; Grimme, S. J. Chem. Theory Comput. 2009, 5, 993.

(23) Gloaguen, E.; de Courcy, B.; Piquemal, J.-P.; Pilmé, J.; Parisel, O.; Pollet, R.; Biswal, H. S.; Piuzzi, F.; Tardivel, B.; Broquier, M.; Mons, M. J. Am. Chem. Soc. **2010**, *132*, 11860.

(24) Scalmani, G; Frisch, M. J. J. Chem. Phys. 2010, 132, 114110.

(25) (a) Suffert, J.; Toussaint, D. J. Org. Chem. 1995, 60, 3550.
(b) Georgy, M.; Boucard, V.; Debleds, O.; Dal Zotto, C.; Campagne, J.-M. Tetrahedron 2009, 65, 1758. (c) Yadav, J. S; Subba Reddy, B. V.; Srinivasa Rao, T.; Bala, B.; Krishna, M.; Narayana Kumar, G. G. K. S. Chem. Lett. 2007, 36, 1472. (d) Shintani, R.; Hayashi, T. Org. Lett. 2005, 7, 2071.

(26) Chen, Y. K.; Yoshida, M.; MacMillan, D. W. C. J. Am. Chem. Soc. 2006, 128, 9328.

(27) Nunno, L.; Di ; Scilimati, A.; Vitale, P. *Tetrahedron* **2005**, *61*, 2623.

(28) Pourzal, A.-A. Synthesis 1983, 717.

(29) Garcia Martinez, A.; Herrera Fernandez, A.; Moreno Jimenez, F. J. Org. Chem. **1992**, 57, 1627.

(30) Schomaker, J. M.; Delia, T. J. J. Org. Chem. 2001, 66, 7125.

(31) Karpov, A. S.; Müller, T. J. J. Synthesis 2003, 2815.